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Wave-Particle Duality and its Implications for General Chemistry Textbooks Mansoor Niaz, Cecilia Marciano, 2012-04-26 It goes without saying that atomic structure including its dual wave particle nature cannot be demonstrated in the classroom Thus for most science teachers especially those in physics and chemistry the textbook is their key resource and their students core source of information Science education historiography recognizes the role played by the history and philosophy of science in developing the content of our textbooks and with this in mind the authors analyze more than 120 general chemistry textbooks published in the USA based on criteria derived from a historical reconstruction of wave particle duality They come to some revealing conclusions including the fact that very few textbooks discussed issues such as the suggestion by both Einstein and de Broglie and before conclusive experimental evidence was available that wave particle duality existed Other large scale omissions included de Broglie s prescription for observing this duality and the importance of the Davisson Germer experiments as well as the struggle to interpret the experimental data they were collecting Also untouched was the background to the role played by Schrödinger in developing de Broglie s ideas The authors argue that rectifying these deficiencies will arouse students curiosity by giving them the opportunity to engage creatively with the content of science curricula They also assert that it isn t just the experimental data in science that matters but the theoretical insights and unwonted inspirations too In addition the controversies and discrepancies in the theoretical and experimental record are key drivers in understanding the development of science as we know it today Evolving Nature of Objectivity in the History of Science and its Implications for Science Education Mansoor Niaz, 2017-10-26 This book explores the evolving nature of objectivity in the history of science and its implications for science education It is generally considered

that objectivity certainty truth universality the scientific method and the accumulation of experimental data characterize both science and science education Such universal values associated with science may be challenged while studying controversies in their original historical context The scientific enterprise is not characterized by objectivity or the scientific method but rather controversies alternative interpretations of data ambiguity and uncertainty Although objectivity is not synonymous with truth or certainty it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific Recent scholarship in history and philosophy of science has shown that it is not the experimental data Baconian orgy of quantification but rather the diversity plurality in a scientific discipline that contributes toward understanding objectivity History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others This book is remarkable for its insistence that the philosophy of science and in particular that discipline s analysis of objectivity as the supposed hallmark of the scientific method is of direct value to teachers of science Meticulously yet in a most readable way Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks it s fascinating how certain perspectives fade while basic questions show no sign of going away There are few books that take both philosophy and education seriously this one does Roald Hoffmann Cornell University chemist writer and Nobel Laureate in Chemistry

Human Chemistry (Volume Two)
 Libb Thims, 2007-09-01 Volume two begins with Goethe s theories of affinities i e the chemical reaction view of human life in 1809 This is followed by the history of how the thermodynamic 1876 and quantum 1905 revolutions modernized chemistry such that affinity the force of reaction is now viewed as a function of thermodynamic free energy reaction spontaneity and quantum valency bond stabilities The composition energetic state dynamics and evolution of the human chemical bond A B is the centerpiece of this process The human bond is what gives yields and takes absorbs energy in life The coupling of this bond energy driven by periodic inputs of solar photons thus triggering activation energies and entropies connected to the dynamical work of life is what quantifies the human reaction process This is followed by topics including mental crystallization template theory LGBT chemistry chemical potential Le Chatelier s principle Muller dispersion forces and human thermodynamics

Chemistry for Cooks Sandra C. Greer, 2023-01-10 A fun approach to teaching science that uses cooking to demonstrate principles of chemistry for undergraduate students who are not science majors high school students culinary students and home cooks How does an armload of groceries turn into a culinary masterpiece In this highly accessible and informative text Sandra C Greer takes students into the kitchen to show how chemistry with a dash of biology and physics explains what happens when we cook Chemistry for Cooks provides all the background material necessary for nonscientists to understand essential chemical processes and to see cooking as an enjoyable application of science Greer

uses a variety of practical examples including recipes to instruct readers on the molecular structure of food the chemical reactions used in cooking to change the nature of food and the essentials of nutrition and taste She also offers kitchen hints and exercises based on the material in each chapter plus do it yourself projects to encourage exploration of the chemistry that takes place when we cook food Features Perfect for science courses aimed at non science majors does not require prior knowledge of chemistry physics or biology Equally useful for general readers home and professional cooks and culinary students Topics include what matter is made of how the structure of matter is altered by heat how we treat food in order to change its microscopic structure why particular procedures or methods are used in the kitchen and how to think critically about various cooking methods A reference section at the end of each chapter points readers to resources for further study Additional online resources include a solutions manual a sample syllabus and PowerPoint slides of all tables and figures

Principles of Qualitative Inorganic Analysis: Precipitation, Separation and Identification of Cations Huda S.

Alhasan,Nadiyah Alahmadi,2021-01-26 This book provides notes for basic laboratory experiments in qualitative analysis of cations The book introduces readers to basic methods and laboratory safety Subsequent chapters cover six groups of cations Each chapter explains important details that are required to understand how a particular analytical method works for detecting cations in samples starting from sedimentation and ending with the identification Key Features Simple reader friendly format introductory notes and summary Covers several groups of metals Appendix for handy reference with tables and references This is a useful textbook for early chemistry students and teachers as it equips the readers with sufficient information required to analyze chemical samples and deduce the presence of specific cations as part of laboratory coursework CRC Handbook of Basic Tables for Chemical Analysis Thomas J. Bruno,Paris D.N. Svoronos,2020-07-30

Researchers in chemistry chemical engineering pharmaceutical science forensics and environmental science make routine use of chemical analysis but the information these researchers need is often scattered in different sources and difficult to access The CRC Handbook of Basic Tables for Chemical Analysis Data Driven Methods and Interpretation Fourth Edition is a one stop reference that presents updated data in a handy format specifically designed for use when reaching a decision point in designing an analysis or interpreting results This new edition offers expanded coverage of calibration and uncertainty and continues to include the critical information scientists rely on to perform accurate analysis Enhancements to the Fourth Edition Compiles a huge array of useful and important data into a single convenient source Explanatory text provides context for data and guidelines on applications Coalesces information from several different fields Provides information on the most useful wet chemistry methods as well as instrumental techniques with an expanded discussion of laboratory safety Contains information of historical importance necessary to interpret the literature and understand current methodology Unmatched in its coverage of the range of information scientists need in the lab this resource will be referred to again and again by practitioners who need quick easy access to the data that forms the basis for experimentation and analysis **General**

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