

# Orbitals and Bonding Test #14 SCH4U Name:

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(Product)  
15 10 8 6

## Assessment of Learning

Multiple Choice: Circle the most correct answer. (15K)

1. After the 5s subshell of an atom is filled with electrons, the next electron added will enter the

- A) 4d subshell B) 5p subshell  
C) 5d subshell D) 4f subshell

2. Which of the following is a correct set of quantum numbers for an electron in a 5f orbital?

- A)  $n = 5, l = 3, m_l = +1$  B)  $n = 5, l = 2, m_l = +3$  C)  $n = 4, l = 3, m_l = 0$   
D)  $n = 4, l = 2, m_l = +1$  E)  $n = 5, l = 4, m_l = 3$

3. Which one of the following combinations of quantum numbers is not allowed?

|    | $n$ | $l$ | $m_l$ |
|----|-----|-----|-------|
| A) | 3   | 0   | 0     |
| B) | 3   | 1   | -1    |
| C) | 3   | 2   | 2     |
| D) | 3   | 2   | -1    |
| E) | 3   | 3   | 2     |

4. The orientation in space of an atomic orbital is associated with

- A) the principal quantum number ( $n$ ). B) the angular momentum quantum number ( $l$ ).  
C) the magnetic quantum number ( $m_l$ ). D) the spin quantum number ( $m_s$ ).  
E) none of the above

5. Consider the following electron configurations to answer the question that follows:

(i)  $1s^2 2s^2 2p^5 3s^1$  (ii)  $1s^2 2s^2 2p^4 3s^2$  (iii)  $1s^2 2s^2 2p^5 3s^2 3p^1$

(iv)  $1s^2 2s^2 2p^5 3s^2 3p^2$  (v)  $1s^2 2s^2 2p^5 3s^2 3p^3$

The electron configuration belonging to the atom with the highest 1st ionization energy is

- A) (i) B) (ii) C) (iii) D) (iv) E) (v)

6. Which atomic orbitals overlap to form the C-O bond of dimethyl ether,  $(CH_3)_2O$ ?

- A) C  $2s^2$  + O  $2s^2$  B) C  $2s^2$  + O  $2p$   
C) C  $2s^2$  + O  $2s^2$  D) C  $2s^2$  + O  $2s^2$

# General Chemistry Chapter 14 Covalent Bonding Orbitals Exam

**A Gutmann**



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web sep 21 2023 tick the disclaimer and click on get aadhaar otp fill the otp received on the registered mobile number and submit the application after submitting form 19 follow similar steps to submit form 10c after checking everything the amount will be deposited in your bank account linked to uan

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web epf form 19 will have to be filled up by an employee holding a provident fund account the form is used only for withdrawing epf amount at the time of retirement or when an employee quits also if an employee exits a job he she will have to be unemployed for at least two months to utilise this form for withdrawal

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web form 10c is a form that must be filled for withdrawal of your pension of the 12 contributed to your epf account 8 33 is directed to pension i e eps account though this amount is secured for your retirement it can be withdrawn during necessary

circumstances like unemployment for 2 or more months medical emergency and others