

HONORS CHEM 1st SEMESTER EXAM STUDY GUIDE FALL 2010

STRATEGY: Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a starting point to identify the areas on which you need to spend more study time. For those areas, go back to homework assignments, quizzes, and reviews to practice more problems. I would also recommend going through all of your tests since these questions are only samples and do not include specific examples of how vocabulary and other conceptual information might appear in a multiple-choice or other format. Remember you can access notes and reviews on my website.

IMPORTANT INFORMATION:

- This must be handed in BEFORE taking your final exam.
- If entirely complete, 5 points will be added to your exam grade
- All work must be shown in DETAIL on lined paper

Introduction to Chemistry – Chapter 1

1. Identify safe/unsafe lab practices.
2. List pieces and uses of lab equipment.
3. List the requirements for tables and graphs.
4. What is plotted on the x-axis and on the y-axis?
5. Make a density graph.
6. What is water displacement?

Matter and Change – Chapter 2

7. Classify the following substances as *solid*, *liquid*, *gas*, or *plasma* based on their properties.
 - a. flexible volume, high KE, particles can disperse freely.
 - b. flexible volume, very high KE, particles are charged.
 - c. fixed volume, very low KE, orderly particles.
 - d. fixed volume, low KE, particles can move past each other.
8. Classify the following as element, compound, heterogeneous mixture, or solution.
 - a. graphite (carbon)
 - b. grape juice
 - c. table salt (NaCl)
 - d. pepper
9. Classify the following as *chemical* or *physical changes*.
 - a. cutting wire
 - b. ripening tomato
 - c. apple slices turning brown
 - d. compressing a gas
10. Classify the following properties as *physical* or *chemical*.
 - a. melts at 68.0°C
 - b. corrosive
 - c. reacts violently with water
 - d. decomposes in air
 - e. magnetic
11. Classify the following properties as *intensive* or *extensive*.
 - a. melting point
 - b. volume
 - c. mass
 - d. reactivity
 - e. density

VOCAB: kinetic molecular theory

General Chemistry 1st Semester Final Exam

ML Morrison

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Table of Contents General Chemistry 1st Semester Final Exam

1. Understanding the eBook General Chemistry 1st Semester Final Exam
 - The Rise of Digital Reading General Chemistry 1st Semester Final Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry 1st Semester Final Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry 1st Semester Final Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry 1st Semester Final Exam
 - Personalized Recommendations
 - General Chemistry 1st Semester Final Exam User Reviews and Ratings
 - General Chemistry 1st Semester Final Exam and Bestseller Lists
5. Accessing General Chemistry 1st Semester Final Exam Free and Paid eBooks
 - General Chemistry 1st Semester Final Exam Public Domain eBooks
 - General Chemistry 1st Semester Final Exam eBook Subscription Services
 - General Chemistry 1st Semester Final Exam Budget-Friendly Options
6. Navigating General Chemistry 1st Semester Final Exam eBook Formats

- ePub, PDF, MOBI, and More
- General Chemistry 1st Semester Final Exam Compatibility with Devices
- General Chemistry 1st Semester Final Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry 1st Semester Final Exam
 - Highlighting and Note-Taking General Chemistry 1st Semester Final Exam
 - Interactive Elements General Chemistry 1st Semester Final Exam
- 8. Staying Engaged with General Chemistry 1st Semester Final Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry 1st Semester Final Exam
- 9. Balancing eBooks and Physical Books General Chemistry 1st Semester Final Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry 1st Semester Final Exam
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine General Chemistry 1st Semester Final Exam
 - Setting Reading Goals General Chemistry 1st Semester Final Exam
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of General Chemistry 1st Semester Final Exam
 - Fact-Checking eBook Content of General Chemistry 1st Semester Final Exam
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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