

Chapter Test A

Chapter: Chemical Equations and Reactions

on this page, indicating which test item is the same as given or has been completed with correction or from answers with questions.

1. The mass number of an atom is equal to the number of protons plus the number of neutrons.
 - a. This number is constant.
 - b. The number of protons and neutrons is constant.
 - c. The number of protons is equal to the number of neutrons.
 - d. None of the above.
2. In the reaction described by a chemical equation,

$$\text{reactants} \rightarrow \text{products}$$
 the number of atoms of each element is the same on both sides of the equation.
 - a. The number of atoms of each element is the same.
 - b. The number of atoms of each element is the same.
 - c. The number of atoms of each element is the same.
 - d. None of the above.
3. Which of the following is a chemical equation?
 - a. The number of atoms of each element is the same on both sides of the equation.
 - b. The number of atoms of each element is the same on both sides of the equation.
 - c. The number of atoms of each element is the same on both sides of the equation.
 - d. The number of atoms of each element is the same on both sides of the equation.
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 - c. The number of atoms of each element is the same on both sides of the equation.
 - d. The number of atoms of each element is the same on both sides of the equation.
8. Which of the following is a chemical equation?
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 - b. The number of atoms of each element is the same on both sides of the equation.
 - c. The number of atoms of each element is the same on both sides of the equation.
 - d. The number of atoms of each element is the same on both sides of the equation.

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Clemens Wendtner



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