

Chemical Bonding

Key Concepts

- Section 1: Introduction to Chemical Bonding**
 What is a chemical bond and why does it form?
- Section 2: Chemical Bonding and Molecular Compounds**
 What is a molecular formula?
 What are the characteristics of a covalent bond?
 How does a Lewis structure represent a molecule?
- Section 3: Ionic Bonding and Ionic Compounds**
 What is a ionic bond?
 How are ionic and molecular compounds different?
- Section 4: Metallic Bonding**
 How does metallic bonding affect the properties of metals?
- Section 5: Molecular Geometry**
 What is VSEPR theory and how is it used?
 What other factors affect the shape of covalent molecules?

Learning Objectives
 LO 1: Understand the formation of chemical bonds and the types of chemical compounds that form.
 LO 2: Understand the formation of molecular compounds and the types of chemical bonds that form.
 LO 3: Understand the formation of ionic compounds and the types of chemical bonds that form.
 LO 4: Understand the formation of metallic compounds and the types of chemical bonds that form.
 LO 5: Understand the formation of covalent compounds and the types of chemical bonds that form.
 LO 6: Understand the formation of ionic compounds and the types of chemical bonds that form.
 LO 7: Understand the formation of metallic compounds and the types of chemical bonds that form.
 LO 8: Understand the formation of covalent compounds and the types of chemical bonds that form.
 LO 9: Understand the formation of ionic compounds and the types of chemical bonds that form.
 LO 10: Understand the formation of metallic compounds and the types of chemical bonds that form.

Review Previous Concepts

1. What are valence electrons?

2. What quantity does the electronegativity of an atom represent?

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Didier Musso



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