

Enthalpy of Formation

- The ΔH°_f for ethanol ($\text{C}_2\text{H}_5\text{OH}$) is depicted in the following thermochemical equation.



$$\Delta H^\circ_f = - 277.7 \text{ kJ/mol}$$

Notice: standard state of C = graphite

H = $\text{H}_2 (\text{g})$

O = $\text{O}_2 (\text{g})$

can use fractional coefficients

Enthalpy Change Of Formation Value For KHCO_3

Yijin Wang



Enthalpy Change Of Formation Value For K_2CO_3 :

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