



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

NATIONAL SENIOR CERTIFICATE (NSC) OCTOBER/NOVEMBER EXAMINATIONS TIMETABLE – 2024 (Amended as at 14 June 2024)

WEEK 1	09:00	14:00
Monday 21/10	English HL P3 (3hrs) English FAL P3 (2½hrs) English SAL P3 (2½hrs)	Afrikaans HL P3 (3hrs) Afrikaans FAL P3 (2½hrs) Afrikaans SAL P3 (2½hrs)
Tuesday 22/10	Computer Applications Tech P1 (3hrs) Practical	Arabic, French, Italian, Mandarin, Modern Greek, Serbian, Spanish SAL P1 (2hrs) Latin SAL P1 (3hrs) Portuguese HL, FAL, SAL P1 (2hrs) Hebrew SAL P1 (2hrs), German HL, SAL P1 (2hrs)
Wednesday 23/10	Business Studies P1 (2hrs)	Information Technology P1 (3hrs) Practical
Thursday 24/10	isiZulu, isiXhosa, siSwati, isiNdebele HL P3 (3hrs), FAL P3 (2½hrs), SAL P3 (2½hrs)	Hindi, Gujarati, Tamil, Telegu, Urdu HL P1 (2hrs) Hindi, Gujarati, Tamil, Telegu, Urdu FAL P1 (2hrs) Hindi, Gujarati, Tamil, Telegu, Urdu SAL P1 (2hrs)
Friday 25/10	Sepedi, Sesotho, Setswana, Xitsonga, Tshivenda HL P3 (3hrs), FAL P3 (2½hrs), SAL P3 (2½hrs) South African Sign Language HL P3 (3hrs)	Arabic, Italian, Mandarin, Modern Greek, Serbian, Spanish SAL P2 (2hrs) Latin SAL P2 (2hrs) Portuguese HL P2 (2½hrs), FAL, SAL P2 (2hrs) Equine Studies (3hrs)
WEEK 2	09:00	14:00
Monday 28/10	English HL P1 (2hrs) English FAL P1 (2hrs) English SAL P1 (2hrs)	Hindi, Gujarati, Tamil, Telegu, Urdu HL P2 (2½hrs) Hindi, Gujarati, Tamil, Telegu, Urdu FAL P2 (2hrs) Hindi, Gujarati, Tamil, Telegu, Urdu SAL P2 (2hrs) Hebrew SAL P2 (2hrs) German HL P2 (2½hrs), SAL P2 (2hrs) Marine Sciences P1 (2½hrs)
Tuesday 29/10	History P1 (3hrs)	Engineering Graphics and Design P1 (3hrs) French SAL P2 (2hrs)
Wednesday 30/10	isiZulu, isiXhosa, siSwati, isiNdebele HL P1 (2hrs), FAL P1 (2hrs), SAL P1 (2hrs)	Business Studies P2 (2hrs)
Thursday 31/10	NON-EXAMINATION DAY	
Friday 01/11	Mathematics P1 (3hrs) Mathematical Literacy P1 (3hrs) Technical Mathematics P1 (3hrs)	Religion Studies P1 (2hrs) Sport and Exercise Science (3hrs)
WEEK 3	09:00	14:00
Monday 04/11	Mathematics P2 (3hrs) Mathematical Literacy P2 (3hrs) Technical Mathematics P2 (3hrs)	Hindi, Gujarati, Tamil, Telegu, Urdu HL P3 (2½hrs) FAL P3 (2½hrs) Portuguese, German HL P3 (2½hrs) Portuguese FAL P3 (2½hrs)
Tuesday 05/11	Geography (Climate and Weather, Geomorphology and Map Work) P1 (3hrs)	Computer Applications Tech P2 (Theory) (3hrs) Nautical Science P1 (3hrs)
Wednesday 06/11	Sepedi, Sesotho, Setswana, Xitsonga, Tshivenda HL P1 (2hrs), FAL P1 (2hrs), SAL P1 (2hrs) South African Sign Language HL P1 (2hrs)	Information Technology P2 (Theory) (3hrs)
Thursday 07/11	History P2 (3hrs)	Consumer Studies (3hrs) Hospitality Studies (3hrs)
Friday 08/11	Physical Sciences (Physics) P1 (3hrs) Technical Sciences P1 (3hrs)	Dramatic Arts (3hrs)

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Table of Contents Matric 2014 Final Examination Timetable For Caps

1. Understanding the eBook Matric 2014 Final Examination Timetable For Caps
 - The Rise of Digital Reading Matric 2014 Final Examination Timetable For Caps
 - Advantages of eBooks Over Traditional Books
2. Identifying Matric 2014 Final Examination Timetable For Caps
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Matric 2014 Final Examination Timetable For Caps
 - User-Friendly Interface
4. Exploring eBook Recommendations from Matric 2014 Final Examination Timetable For Caps
 - Personalized Recommendations
 - Matric 2014 Final Examination Timetable For Caps User Reviews and Ratings
 - Matric 2014 Final Examination Timetable For Caps and Bestseller Lists
5. Accessing Matric 2014 Final Examination Timetable For Caps Free and Paid eBooks
 - Matric 2014 Final Examination Timetable For Caps Public Domain eBooks
 - Matric 2014 Final Examination Timetable For Caps eBook Subscription Services
 - Matric 2014 Final Examination Timetable For Caps Budget-Friendly Options

6. Navigating Matric 2014 Final Examination Timetable For Caps eBook Formats
 - ePub, PDF, MOBI, and More
 - Matric 2014 Final Examination Timetable For Caps Compatibility with Devices
 - Matric 2014 Final Examination Timetable For Caps Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Matric 2014 Final Examination Timetable For Caps
 - Highlighting and Note-Taking Matric 2014 Final Examination Timetable For Caps
 - Interactive Elements Matric 2014 Final Examination Timetable For Caps
8. Staying Engaged with Matric 2014 Final Examination Timetable For Caps
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Matric 2014 Final Examination Timetable For Caps
9. Balancing eBooks and Physical Books Matric 2014 Final Examination Timetable For Caps
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matric 2014 Final Examination Timetable For Caps
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Matric 2014 Final Examination Timetable For Caps
 - Setting Reading Goals Matric 2014 Final Examination Timetable For Caps
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Matric 2014 Final Examination Timetable For Caps
 - Fact-Checking eBook Content of Matric 2014 Final Examination Timetable For Caps
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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- Interactive and Gamified eBooks

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web these structures also known as lewis structures or electron dot structures are drawings that visually demonstrate how electrons are shared and arranged around atoms the electrons denoted as dots are called lone pairs and belong to an individual atom electrons denoted as lines are bonds and show the sharing of two electrons between two

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web when you are trying to draw a lewis structure there are guidelines to do it number of steps can be changed according to the complexity of the molecule or ion because p2o5 molecule is a bit complex molecule if you are a beginner to lewis structure drawing you have to be careful when it starts to draw

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web jun 27 2022 a lewis electron dot diagram or electron dot diagram or a lewis diagram or a lewis structure is a representation of the valence electrons of an atom that uses dots around the symbol of the element the number of dots equals the number of valence electrons in the atom

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web oct 8 2023 introduction lewis structure is basically a graphic representation of the electron distribution around an atom the major reason why learning lewis dot structure is important is that it helps in predicting the number and type of bonds which can be formed around an atom it also helps in predicting the geometry of the molecule