



# Hp 200 5020 Desktops Owners Manual

**Sebastian Brünink**



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web we utilize the equation that relate amount remaining initial mass and number of half lives  $n = \frac{t}{t_{1/2}} \times \log_2 \frac{N_0}{N}$  125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore  $n = \frac{t}{t_{1/2}} \times \log_2 \frac{N_0}{N}$  now using the half life equation in example 1 we have

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