

**Chapter 17: Thermochemistry***Section 1: The Flow of Energy – Heat and Work*

1. What is thermochemistry? **The study of the energy changes that occur in changes of state or during chemical reactions.**
2. Heat always flows from a **warmer** object to a **cooler** object.
3. Given that  $1 \text{ cal} = 4.184 \text{ J}$ :
  - a. How many calories are in  $325.9 \text{ J}$ ? **77.89 cal**
  - b. How many J are in  $34.7 \text{ cal}$ ? **145.18 J**
4. What is the Law of Conservation of Energy (aka the First Law of Thermodynamics)? **States that in chemical and physical processes, energy is neither created nor destroyed – it is transferred or conserved.**
5. In an exothermic reaction:
  - a. Heat is **released** / absorbed
  - b.  $q$  is **positive** / **negative**
  - c.  $\Delta H$  is **positive** / **negative**
  - d. Heat is a **reactant** / **product**
6. In an endothermic reaction:
  - a. Heat is **released** / **absorbed**
  - b.  $q$  is **positive** / **negative**
  - c.  $\Delta H$  is **positive** / **negative**
  - d. Heat is a **reactant** / **product**
7. Diagram **A** represents an exothermic reaction, whereas diagram **B** represents an endothermic reaction:
8. What is the difference between heat capacity and specific heat capacity? **Specific heat is the amount of energy to increase the temperature of 1g of substance by  $1^\circ\text{C}$**
9. The temperature of a  $34.1 \text{ g}$  sample of a substance increases from  $29^\circ\text{C}$  to  $51^\circ\text{C}$  when the sample absorbs  $912 \text{ J}$  of heat. What is the specific heat of the substance?  **$1.126 \text{ J/g}^\circ\text{C}$**
10. When  $867 \text{ J}$  of heat are added to  $9.3 \text{ g}$  of a substance at STP, the temperature increases from  $0^\circ\text{C}$  to  $14^\circ\text{C}$ . What is the specific heat of the substance?  **$6.66 \text{ J/g}^\circ\text{C}$**

# Holt Modern Chemistry Chapter 17 Review Answers

**Jiyuan Zhang**



## **Holt Modern Chemistry Chapter 17 Review Answers:**

## Reviewing **Holt Modern Chemistry Chapter 17 Review Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Holt Modern Chemistry Chapter 17 Review Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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