

## **Report FOR EXPERIMENT # 1: DETERMINATION OF MELTING POINTS**

Your name

TA's name

Your Partner's name

Lab Section

### **RESULTS:**

Melting point of urea: 134-154 °C

Melting point of trans-cinnamic acid: 133-134 °C

Melting range of a mixture of urea and trans-cinnamic acid: 110-125 °C

Melting point of unknown: 133-133.5 °C

Melting point of unknown mixed with urea: 112-123 °C

Melting point of unknown mixed with Melting point of unknown: 133-133.5 °C

Identity of unknown: Melting point of unknown: 133-133.5 °C

### **CONCLUSION:**

In this lab we used melting points to determine the purity of organic compounds. Pure compounds were seen to have a narrow melting range, while impurities were seen to cause the melting range to become lower and broader. From the mixture melting points we were definitively able to identify our unknown as trans-cinnamic acid.

### **QUESTIONS: DETERMINATION OF MELTING POINTS**

1. What is a safety hazard associated with this lab session?

*One will be using instruments that will be heated to high temperatures. So, one should take precautions when handling these instruments.*

# Lab Report On Melting Points

**Liying Dong**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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