

# Empirical and Molecular Formulas Practice

Read the question carefully for the data provided. Answers must be shown for credit.

Find empirical formula:

1. A compound is found to contain 40.0 g of carbon and 6.67 g of hydrogen. Find the empirical formula.

$$\text{C}_{40.0} \text{H}_{6.67} \times \frac{1 \text{ mole}}{12.01 \text{ g C}} = 3.33 \text{ moles C}$$

$$\text{H}_{6.67} \times \frac{1 \text{ mole}}{1.008 \text{ g H}} = 6.62 \text{ moles H}$$

C<sub>3.33</sub>H<sub>6.62</sub>

2. A compound is found to contain 50.0 g of carbon, 4.31 g of hydrogen, and 10.67 g of oxygen. Find the empirical formula.

$$\text{C}_{50.0} \text{H}_{4.31} \text{O}_{10.67} \times \frac{1 \text{ mole}}{12.01 \text{ g C}} = 4.16 \text{ moles C}$$

$$\text{H}_{4.31} \times \frac{1 \text{ mole}}{1.008 \text{ g H}} = 4.28 \text{ moles H}$$



$$\text{O}_{10.67} \times \frac{1 \text{ mole}}{16.00 \text{ g O}} = 0.667 \text{ moles O} \quad \text{H}_2\text{C}_2\text{O}_4$$

3. A 10.0 g sample of a compound, a possible alkane, is found to contain 8.33 g of carbon, 1.67 g of hydrogen, and 0.00 g of oxygen. Find the empirical formula.

$$\text{C}_{8.33} \text{H}_{1.67} \times \frac{1 \text{ mole}}{12.01 \text{ g C}} = 0.693 \text{ moles C}$$



$$\text{H}_{1.67} \times \frac{1 \text{ mole}}{1.008 \text{ g H}} = 1.66 \text{ moles H} \quad \text{C}_2\text{H}_5$$

$$\text{C}_{0.693} \text{H}_{1.66} \times \frac{1 \text{ mole}}{1.008 \text{ g H}} = 1.66 \text{ moles H} \quad \text{C}_2\text{H}_5$$

4. A 20.00 g sample of an unknown compound contains 12.00 g carbon, 2.00 g hydrogen, and 6.00 g oxygen. Find the empirical formula.

$$\text{C}_{12.00} \text{H}_{2.00} \text{O}_{6.00} \times \frac{1 \text{ mole}}{12.01 \text{ g C}} = 1.00 \text{ moles C}$$



$$\text{H}_{2.00} \times \frac{1 \text{ mole}}{1.008 \text{ g H}} = 1.98 \text{ moles H}$$

$$\text{O}_{6.00} \times \frac{1 \text{ mole}}{16.00 \text{ g O}} = 0.375 \text{ moles O} \quad \text{H}_2\text{C}_2\text{O}_4$$

# Empirical Formula Pogil Answers

**Jiyuan Zhang**



## **Empirical Formula Pogil Answers:**

## Unveiling the Magic of Words: A Review of "**Empirical Formula Pogil Answers**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Empirical Formula Pogil Answers**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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