Experiment 7 Report Sheet

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			具 医甲基

Land whether the Car		Empirical Formulas
Infaction whether the following data are for Part B, p	in c	Denk No.
	or Part D.	Identify the original sample
 Mass of crucible and lid (g) 	Trial I	Trial 2
2. Mass of crucible, lid, and sample (g)	-25_636	_ 40.347
3. Mass of sample (g) D. 150	-25.786	_ 40.542
4. Instructor's approval of flame and apparatus	_0.150	0.195
5. Mass of crucible, lid, and product		
1st mass measurement (g)	0-0-0	
2nd mass measurement (g)	25.858	_ 40.654
	25.856	40.652
3rd mass measurement (g)	25.859	40.656
6. Final mass of crucible, lid, and product (g)	25.860	40.655
7. Mass of product (g)	0.2249	0.1139
S. Part B. Combination Reaction of Magnesium ar	ad Oxygen	
a. Mass of O (g)	0.014a	
h. Mass ratio of Mg to O	2:1	
c. Moles of Mg (mol)	0.00617 mot	
d. Moles of O (mol)	0.00463 mol	
e. Mole ratio of Mg to O	1:1	
Consensus empirical formula of magnesium oxid	6	1/Mg0
Percent by mass (%): 67.0 % Mg; 3		
art C. Decomposition Reaction of a Pure Compo	ound	
Mass of product = mass CaO (g)	0.224	The second secon
Mass of CO ₂ (g)	0.1263	to the second second

Empirical Formula Of A Compound Lab Report

Thirumalaisamy P. Velavancorresponding

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