

Summary Report

Create a table for each cation studied during this lab (include results for mixtures of cations when appropriate). Each table should include the tests performed, observations and chemical reactions. The following table for Pb^{2+} should be used as a guide.

Cation	Test	Observations	Chemical Reaction
Pb^{2+}	HCl	Formation of a white precipitate	$\text{Pb}^{2+}_{(aq)} + 2\text{Cl}^{-}_{(aq)} \rightarrow \text{PbCl}_{2(s)}$
	Hot water bath		
	K_2CrO_4		

Lab Manual For Chem 1212l

Michael Seilmaier



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