

19.2

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Identify a conjugate acid-base pair in chemical reactions involving acids and bases.
2. Calculate the pH of a solution from the concentration of H^+ or H_3O^+ ions, and vice versa.
3. Determine the pH of a solution from K_a and K_b .

Vocabulary

acid	1. conjugate base	acid solution
acid-base indicator	2. hydronium	acidic solution
acidic solution	3. autoionization	pH

Key Equation

- a. $pH = -\log [H^+]$ (ppm 1.4.1 and 1.4.2)
- b. $[H^+] = 10^{-pH}$ (ppm 1.4.1 and 1.4.2)

Part A Completion

Write the answers to the following questions in the spaces provided. Show necessary calculations and units for calculations involving either mass, volume, or moles.

1. What is the pH of a 0.0010 M HCl solution? a. 0.0010
2. Calculate the pH of the 0.0010 M HCl solution. b. 0.0010
3. What is the pH of a 0.0010 M HCl solution? c. 0.0010
4. What is the pH of a 0.0010 M HCl solution? d. 0.0010
5. What is the pH of a 0.0010 M HCl solution? e. 0.0010
6. What is the pH of a 0.0010 M HCl solution? f. 0.0010
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13. What is the pH of a 0.0010 M HCl solution? m. 0.0010
14. What is the pH of a 0.0010 M HCl solution? n. 0.0010
15. What is the pH of a 0.0010 M HCl solution? o. 0.0010
16. What is the pH of a 0.0010 M HCl solution? p. 0.0010
17. What is the pH of a 0.0010 M HCl solution? q. 0.0010
18. What is the pH of a 0.0010 M HCl solution? r. 0.0010
19. What is the pH of a 0.0010 M HCl solution? s. 0.0010
20. What is the pH of a 0.0010 M HCl solution? t. 0.0010
21. What is the pH of a 0.0010 M HCl solution? u. 0.0010
22. What is the pH of a 0.0010 M HCl solution? v. 0.0010
23. What is the pH of a 0.0010 M HCl solution? w. 0.0010
24. What is the pH of a 0.0010 M HCl solution? x. 0.0010
25. What is the pH of a 0.0010 M HCl solution? y. 0.0010
26. What is the pH of a 0.0010 M HCl solution? z. 0.0010

Hydrogen Ions And Acidity Section Review Answers

Minjie Lin



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