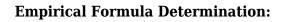
# How to Calculate an Empirical Formula

A 283.9 g sample of unknown composition was carefully analyzed and found to consist of 49.8 g of sodium, 113.6 g of chromium, and 121.4 g of oxygen.

Determine the empirical formula of the unknown compound.

# **Empirical Formula Determination**

**RJ Shavelson** 



This book delves into Empirical Formula Determination. Empirical Formula Determination is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Empirical Formula Determination, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Empirical Formula Determination
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  - Chapter 3: Empirical Formula Determination in Everyday Life
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  - ∘ Chapter 5: Conclusion
- 2. In chapter 1, this book will provide an overview of Empirical Formula Determination. This chapter will explore what Empirical Formula Determination is, why Empirical Formula Determination is vital, and how to effectively learn about Empirical Formula Determination.
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- 6. In chapter 5, the author will draw a conclusion about Empirical Formula Determination. This chapter will summarize the key points that have been discussed throughout the book.
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