





# Empirical Formula

## Empirical Formula

A formula that gives the simplest whole-number ratio of the atoms of each element in a compound.

| Molecular Formula  | Empirical Formula     |
|--|-----------------------|
| $\text{H}_2\text{O}_2$                                       | HO                    |
| $\text{C}_6\text{H}_{12}\text{O}_6$                          | $\text{CH}_2\text{O}$ |
| $\text{CH}_3\text{O}$                                      | $\text{CH}_3\text{O}$ |
| $\text{CH}_3\text{OOCH} = \text{C}_2\text{H}_4\text{O}_2$  | $\text{CH}_2\text{O}$ |

# Empirical Formula And Molecular Formula Pogil

**LP Steffe**



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