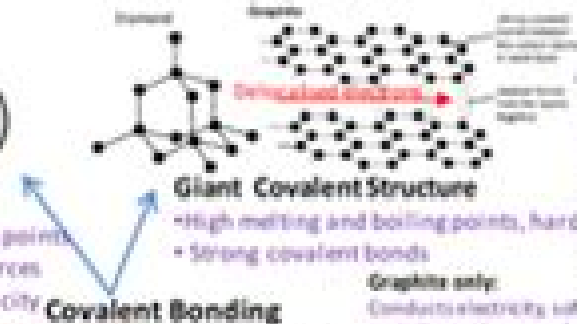
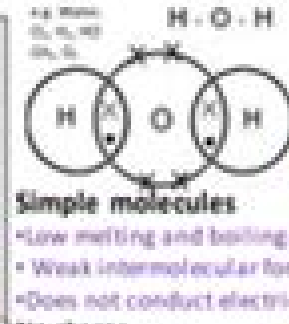
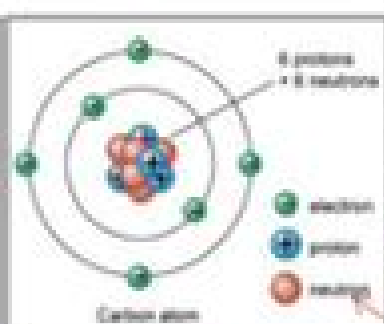




Periodic Table of the Elements

Legend:

- Hydrogen
- Alkali metals
- Alkaline earth metals
- Transition metals
- Alkali earth metals
- Noble gases
- Halogens
- Lanthanide/actinide series

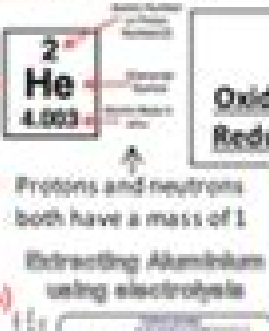
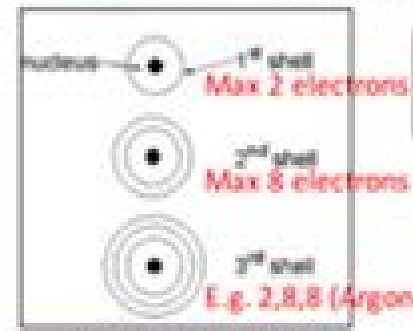
The periodic table displays elements from Hydrogen (H) to Oganesson (Og), with color-coded groups as defined in the legend.

An atom:
 Number of Protons = Number of Electrons
 Mass number = Protons + Neutrons

C2 Chemistry

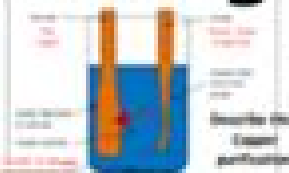
Periodic table arranges elements by
ATOMIC NUMBER (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
 E.g. Lithium 2,1 is a group 1 element.

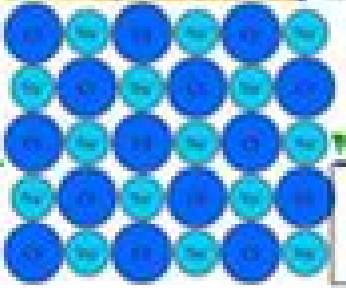
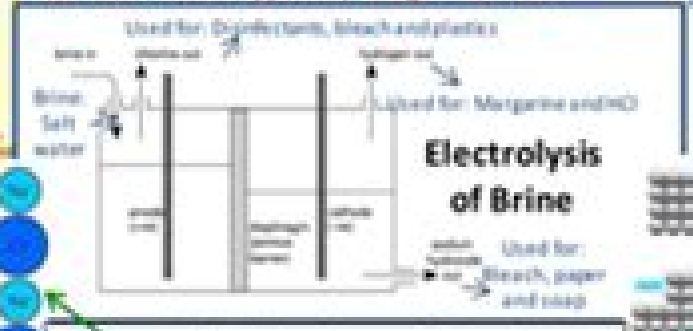
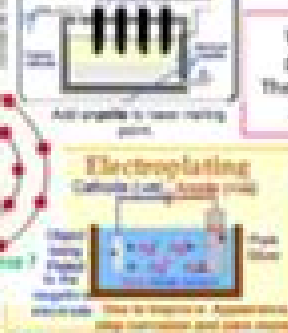
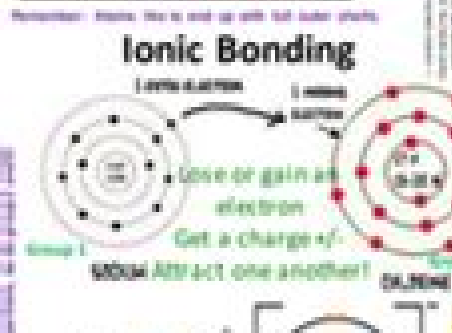
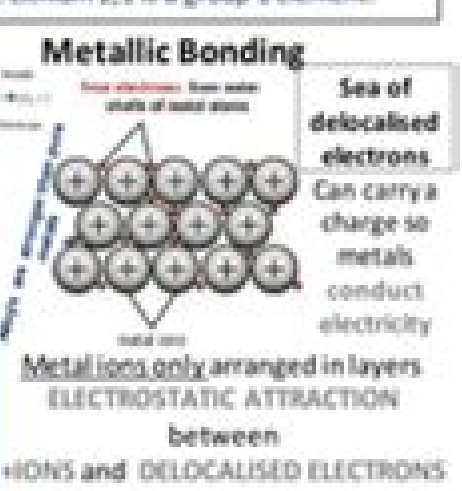
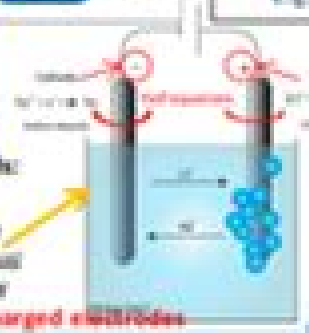


OIL RIG

Oxidation is loss (of electrons)
 Reduction is gain (of electrons)



Electrolysis:
 Splitting up a substance using electricity



NANO-TECHNOLOGY
 Means **REALLY REALLY REALLY** small!!!!

May 2013 Chemistry C2

Hussin A.Rothana



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