

KENDRIYA VIDYALAYA SANGATHAN
LUCKNOW REGION
PRE BOARD EXAMINATION 2020-21
CLASS- XII
Subject – Chemistry

Time allowed : 3 Hrs

Maximum Marks : 70

General Instructions :-

1. There are 33 questions in this question paper. All questions are compulsory.
2. Section A : Q.No. 1 to 2 are case-based questions having four MCQs or Reason Assertion type based Given passage each carrying 1 mark.
 Q. No. 3 to 16 are MCQs and Reason Assertion type questions carrying 1 mark each.
3. Section C : Q. No. 17 to 25 are short answer questions and carry 2 marks each.
4. Section D : Q. No. 26 to 30 are short answer questions and carry 3 marks each.
5. Section E : Q. No. 31 to 33 are long answer questions carrying 5 marks each.
6. There is no overall choice. However, internal choices have been provided.
7. Use of calculators and log tables is not permitted.

SECTION A (OBJECTIVE TYPE)

1	<p>Read the passage given below and answer the following questions: The d- block of the periodic table contains the elements of the groups 3-12 and are known as transition elements. In general, the electronic configuration of these elements is $(n-1)d^{1-10}ns^{1-2}$. The d orbitals of the penultimate energy level in their atoms receive electrons giving rise to the three rows of the transition metals i.e. 3d, 4d and 5d series. However, Zn, Cd and Hg are not regarded as transition elements. Transition elements exhibit certain characteristic properties like variable oxidation states, complex formation, formation of coloured ions and alloys, catalytic activity etc. Transition metals are hard (except Zn, Cd and Hg) and have a high melting point. The following questions are multiple choice questions. Choose the most appropriate answer:</p> <p>(i) Which transition metal of 3d series does not show variable oxidation states? (a) Cr (b) Ti (c) Sc (d) Fe</p> <p>(ii) Which of the following statement about transition element is not correct? (a) They show variable oxidation states. (b) They exhibit diamagnetic and paramagnetic properties. (c) All the ions are coloured. (d) They exhibit catalytic properties.</p> <p>(iii) The magnetic moment is associated with its spin angular momentum and orbital angular momentum. Spin only magnetic moment of Cr^{3+} ion is (a) 2.87 BM (b) 3.87 BM (c) 3.47 BM (d) 3.57 BM</p> <p>(iv) Which of the following has maximum number of unpaired d- electrons? (a) Mg^{2+} (b) Ti^{3+} (c) V^{2+}</p>	4
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