

can be used to deter-

lithium?  $7 \times 14 \rightarrow 50$

$0.25)(7.02 \text{ u}) \rightarrow 13$

$0.75)(7.02 \text{ u}) \times 6 \rightarrow 50$

$(7.02 \text{ u}) \times 13$

$(7.02 \text{ u}) \times 9.25 \times (7.02)$

chlorine to form an ionic  
formula  $\text{XCl}_2$ . To which  
table could element X

(3) Group 13

(4) Group 15

$\text{I}^-$

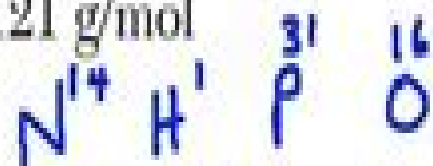
37 What is the gram-formula mass of  $(\text{NH}_4)_3\text{PO}_4$ ?

(1) 112 g/mol

(3) 149 g/mol

(2) 121 g/mol

(4) 242 g/mol



38 In the ground state, which atom has a completely  
filled valence electron shell?

(1) C

(3) Ne

(2) V

(4) Sb

$3\text{N}: 3 \times 14 = 42$

$12\text{H}: 12 \times 1 = 12$

$1\text{P}: 1 \times 31 = 31$

$4\text{O}: 4 \times 16 = 64$

$149$

# June 2012 Chemistry Regents Answers

**K Payea**



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