

১৫। $x^2 - \sqrt{5}x + 1 = 0$ হলে $x + \frac{1}{x} =$ কত?

- (ক) $-2\sqrt{5}$ (খ) $-\sqrt{5}$
(গ) $\sqrt{5}$ (ঘ) $5\sqrt{5}$

১৬। $x + y = 6$ এবং $x - y = 4$ হলে $5xy$ এর মান কত?

- (ক) 16 (খ) 24
(গ) 25 (ঘ) 36

১৭। $U = \{1, 2, 3, 4, 5, 6, 7\}$, $A = \{3, 5, 6, 7\}$ হলে A র সমান কত হবে?

- (ক) $\{1, 2, 4\}$ (খ) $\{2, 4, 5\}$
(গ) $\{2, 4, 6\}$ (ঘ) $\{3, 5, 6, 7\}$

১৮। $A = \{3, 4, 5\}$ এবং $B = \{4, 5, 7\}$ হলে—

- i. $A \cup B = \{3, 4, 5, 7\}$
ii. $A \cap B = \{4, 5\}$
iii. $A \subset B$

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) i ও iii
(গ) ii ও iii (ঘ) i, ii ও iii

১৯। $x^2 - x - 42$ এর উৎপাদকে বিশ্লেষিত রূপ নিচের কোনটি?

- (ক) $(x-6)(x-7)$
(খ) $(x-6)(x+7)$
(গ) $(x+6)(x+7)$
(ঘ) $(x-7)(x+6)$

২০। কোনো বিন্দুটি x -অক্ষের উপর অবস্থিত?

- (ক) $(0, -4)$ (খ) $(-4, 0)$
(গ) $(0, 3)$ (ঘ) $(3, -1)$

২১। কোন প্রকৃত ভগ্নাংশের লব ও হরের যোগফল 14 এবং বিয়োজক 8 হলে, ভগ্নাংশটি নিচের কোনটি?

- (ক) $\frac{5}{9}$ (খ) $\frac{9}{5}$
(গ) $\frac{3}{11}$ (ঘ) $\frac{11}{3}$

২২। $m^2 - 6m - 1 = 0$ হলে $m^2 + \frac{1}{m^2} =$ কত?

- (ক) 32
(খ) 34
(গ) 36
(ঘ) 38

২৩। একটি সমকোণী ত্রিভুজের বাহুদ্বয়ের ব্যাসার্ধ 4 সে.মি. এবং উচ্চতা 6 সে.মি.। ত্রিভুজটির বহু পৃষ্ঠের ক্ষেত্রফল কত? ($\pi = 3.14$)

- (ক) 75.36 বর্গ সে.মি.
(খ) 150.72 বর্গ সে.মি.
(গ) 226.08 বর্গ সে.মি.
(ঘ) 301.44 বর্গ সে.মি.

নিচের চিত্র অনুসারে ২৪ ও ২৫নং প্রশ্নের উত্তর দাও।



ABCD বর্জাক্ষেত্রের কর্ণদ্বয় 'O' বিন্দুতে মিলিত হয়েছে এবং $BC = 6$ cm.

২৪। $\angle AOB =$ কত?

- (ক) 30° (খ) 45°
(গ) 60° (ঘ) 90°

২৫। AC এর দৈর্ঘ্য কত?

- (ক) 6 সে.মি. (খ) $6\sqrt{2}$ সে.মি.
(গ) $9\sqrt{2}$ সে.মি. (ঘ) 12 সে.মি.

২৬। যে চতুর্ভুজের দুই জোড়া সম্মিলিত বাহু সমান তাকে কী বলে?

- (ক) অয়ত (খ) বর্গ
(গ) ঘূর্ণি (ঘ) সামান্তরিক

২৭।



ABC ত্রিভুজের ক্ষেত্রফল কত?

- (ক) 24 বর্গ সে.মি. (খ) 36 বর্গ সে.মি.
(গ) 48 বর্গ সে.মি. (ঘ) 60 বর্গ সে.মি.

২৮। একটি ত্রিভুজের বাহুদ্বয়ের অনুপাত $x : x : x\sqrt{2}$ হলে, এর বৃহত্তম কোণটির মান কত?

- (ক) 80° (খ) 85°
(গ) 90° (ঘ) 120°

২৯। চতুর্ভুজের চারটি কোণের সমষ্টি কত?

- (ক) 360° (খ) 300°
(গ) 270° (ঘ) 180°

৩০। একটি ট্র্যাপিজিয়ামের সমান্তরাল বাহু দুইটির দৈর্ঘ্য যথাক্রমে 13 সে.মি. এবং 8 সে.মি.। সমান্তরাল বাহুদ্বয়ের লম্ব দূরত্ব 4 সে.মি. হলে ট্র্যাপিজিয়ামটির ক্ষেত্রফল কত বর্গ?

- (ক) 42 বর্গ সে.মি. (খ) 52 বর্গ সে.মি.
(গ) 84 বর্গ সে.মি. (ঘ) 104 বর্গ সে.মি.

Jsc 201math Question

Daniel F McAuley



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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and Δh is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web pertanyaan serupa iklan perhatikan diagram energi potensial berikut pada diagram di atas energi aktivasi untuk reaksi baliknya ditandai dengan

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web answer see below explanation the rate of a chemical reaction $a + b \rightarrow c + d$ is given in first approximation by $v = k_v a^m b^n$ where k_v k_e e_a r where k depends by many factors as solvent geometry type of reaction e_a is the activation energy r is the universal constant of gases t is the absolute temperature

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web mar 17 2016 3 what is the activation energy of a reaction and how is this energy related to the activated complex of the reaction e_a is the minimum amount of energy for a reaction to occur it is the amount of energy required to create an activated complex 4 what happens when a catalyst is used in a reaction a catalyst changes the reaction

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy

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web key points potential energy is energy that has the potential to become another form of energy an object's potential energy depends on its physical properties and position in a system potential energy comes in many forms such as

gravitational potential energy due to an object's mass and position in a gravitational field

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate ΔH for the reaction and label it with the correct value see attached image

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web aug 16 2019 1 answer sorted by 4 the activation energy is not the average kinetic energy that the colliding particles need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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web all questions are answered herein chemistry tutorial 9 02b potential energy diagrams how to draw and label pe diagrams for exothermic and endothermic reactions and what effect adding a catalyst or inhibitor has on the diagram

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web 1 potential energy diagrams are graphs energy is on the y axis and the x axis indicates geometry energy is a function of geometry 2 the geometry changes refer to changes in bonding changes in atom positions each point on the diagram has the same molecular formula same atoms and electrons the bonding can change however

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web may 1 2013 a potential energy diagram shows the change in potential energy of a system as reactants are converted into products figure below shows basic potential energy diagrams for an endothermic a and an exothermic b reaction recall that the enthalpy change ΔH is positive for an endothermic reaction and negative for an

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web 1 draw a potential energy diagram for an endothermic reaction 2 label your drawing with the following letters a pe of products d energy of activation b pe of reactants e heat of reaction c pe of the activated complex potential energy diagrams time ms 0 00 1 00 2 00 3 00 4 00 5 00 6 00 7 00 8 00 9 00 10 0 potential energy kcal 0 00 10 0

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web use the potential energy diagram to answer the questions below 1 is the overall reaction as shown exothermic or endothermic 2 what is the activation energy for the forward reaction 3 what is the