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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23  $\text{Fe}^{2+} + \text{CO} \rightarrow \text{CO}_2 + \text{Fe}$   
oxidation  $\text{CO} \rightarrow \text{CO}_2 + 2\text{e}^-$  reduction  $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$  24  $3\text{Ag} + \text{Ni}^{3+} \rightarrow 3\text{Ag} + \text{Ni}$  oxidation reduction 25  $\text{Cu} + \text{Pb}^{2+} \rightarrow \text{Pb} + \text{Cu}$  oxidation reduction 26  $\text{O}_2 \rightarrow 2\text{O}$

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