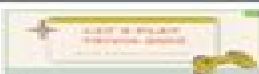
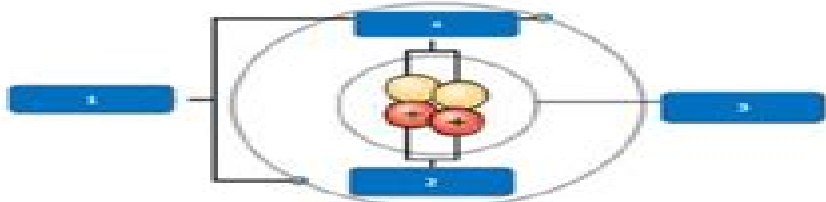
 <div>DAILY LESSON LOG</div>	<div>School</div> Batasan Hills National High School	<div>Grade Level</div> 8	<div>Learning Area</div> Chemistry																				
<div>Teacher</div> MARIA RIZZA E. LLORENTE	<div>Teaching Date and Time</div> for 50 minutes	<div>Quarter</div> 3rd	<div>Day</div> Thursday-Friday																				
<div>DLL#68</div>	<div>Date: March 23-24, 2023</div>																						
<div>I. OBJECTIVE:</div>																							
<div>A. Content Standard</div>	The learners demonstrate understanding of the identity of a substance according to its atomic number																						
<div>B. Performance Standard</div>	The learners should be able to present how water behaves in its different states within the water cycle.																						
<div>C. Learning Competencies</div>	Determine the number of protons, neutrons, and electrons in a particular atom; (S8MT-IIIc-f-10)																						
<div>Specific Objectives:</div>	At the end of the lesson, the students should be able to: 1. Describe the atom and its sub-atomic particles; 2. Determine the number of electrons, protons, and neutrons in an atom; and 3. Explain how atoms benefit life.																						
<div>II. CONTENT</div>	<div>SUBATOMIC PARTICLES</div> <div>Determining the number of protons, electrons and neutrons in an atom</div>																						
<div>III. LEARNING RESOURCES</div>																							
<div>A. References</div>																							
<div>B. Other Learning Resources</div>	<ul style="list-style-type: none">• http://chemed.chem.purdue.edu/genchem/topicreview/bp/ch6/atom_emfame.html• http://www.kentchemistry.com/links/AtomicStructure/protonsws1.htm• https://education2eb.org/lesson_number.html• https://www.nap.edu/read/4818/chapter/2• https://www.youtube.com/watch?v=1M4jysdgmRg																						
<div>IV. PROCEDURE</div>																							
<div>A. Reviewing previous lesson or presenting the new lesson.</div>	<div>ELICIT: TRIVIA QUIZ GAME</div> <div>The class will play trivia quiz game about atoms and molecules</div> <div></div> <div>https://www.canva.com/design/DAP6p7UYH0Q8kUQleQA8isLMxgQK1CZueA/edit</div>																						
<div>B. Establishing a purpose for the lesson</div>	<div>ENGAGE: LABEL THE PICTURE: What's Inside the Atom?</div> <div></div> <div>a. Ask the class to label the picture. b. Introduce subatomic particles. c. Recall the proponent/discovery of the atomic model on the picture.</div> <div>Answer: 1. Electron 2. Proton 3. Nucleus 4. neutron</div>																						
<div>C. Presenting examples/instances of the new lesson</div>	<div>ENGAGE: TABLE ANALYSIS</div> <div>The students will study the table below and answer the questions about Properties of Subatomic Particles.</div> <table><tr><th>Subatomic particle</th><th>Symbol</th><th>Relative charge</th><th>Location</th><th>Mass</th></tr><tr><td>Proton</td><td>p</td><td>+1</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Neutron</td><td>n</td><td>0</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Electron</td><td>e⁻</td><td>-1</td><td>Outside the nucleus</td><td>0.0005 (-1/1836)</td></tr></table>			Subatomic particle	Symbol	Relative charge	Location	Mass	Proton	p	+1	Inside the nucleus	1	Neutron	n	0	Inside the nucleus	1	Electron	e ⁻	-1	Outside the nucleus	0.0005 (-1/1836)
Subatomic particle	Symbol	Relative charge	Location	Mass																			
Proton	p	+1	Inside the nucleus	1																			
Neutron	n	0	Inside the nucleus	1																			
Electron	e ⁻	-1	Outside the nucleus	0.0005 (-1/1836)																			

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**ONS,June Eilers,Martha
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