Six Types of Chemical Reaction Worksheet

Balance the following reactions and indicate which of the six types of chemical reaction are being represented:

NaBr + ___ Ca(OH)₂ → __ CaBr₂ + ___ NaOH

Type of reaction: _____

2) ____ NH₃+ ____ H₂SO₄ → ____ (NH₄)₂SO₄

Type of reaction:

3) ____ C₅H₉O + ____ O₂ → ____ CO₂ + ____ H₂O

Type of reaction: _____

4) Pb + H₃PO₄ → H₂ + Pb₃(PO₄)₂

Type of reaction:

5) ____ Li₃N + ___ NH₄NO₃ → ___ LiNO₃ + ___ (NH₄)₃N

Type of reaction:

6) ____ HBr + ___ Al(OH)₃ → ___ H₂O + ___ AlBr₃

Type of reaction:

7) What's the main difference between a double displacement reaction and an acid-base reaction?

8) Combustion reactions always result in the formation of water. What other types of chemical reaction may result in the formation of water? Write examples of these reactions on the opposite side of this paper.

Lesson 36 Reaction Rates Answers Key

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