LESSON ACTIVITY.

Pass the Proton Acid-Base Theories

Name		
Date	Period	

Purpose

To compare acid and base solutions at the particulate level.

Materials

Acid-Base Solution cards:

Procedure and Questions

1. Sort the cards into three groups based on common features. Write the solution card number, molarity, chemical formula, and chemical name for each solution. Group I. Group 2: Group 3:

		200.000.000.000
4: CHOOH	2: HCL	3: HCL
1: HCL	8: CH2:OH	10: NaOH
5: CH2COOH		
	All have 0.005 M	12: HOH
6: NaCl	POLITICATE UL GUID INI	

9: NaOH All have 0.001 M. 7: CH2OH

11: NH4.

All have 0.010 M

- 2. What do the solutions in each of the three groups have in common?
 - Group 1: same molarity, 0.010 M and a lot of protons
 - Group 2: same molarity, 0.005M and little protons
 - Group 3: same molarity, 0.001 M and lack of protons
- 3. Is there a second way you might sort the cards? If so, explain how.

You can sort them by the patterns/shape of protons

4. Fill in the table with the chemical formula of the substance dissolved and the ions in the solution.

Lesson 12 Unit 4 Toxins Worksheet

Ian Pickup

Lesson 12 Unit 4 Toxins Worksheet:

Resources in Education ,1991

Resources in Education ,1991

The Wall Street Journal ,1986

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