



Discussion Questions

- (i) Explain why representing the ionic reaction between ferric and thiocyanate ions as given in the text viz.



is more appropriate in the following form ?



- (ii) Does the constancy in colour intensity indicate the dynamic nature of equilibrium? Explain your answer with appropriate reasons.
- (iii) What is equilibrium constant and how does it differ from the rate constant?
- (iv) It is always advisable to carry out the present experiment with dilute solutions. Why?
- (v) What will be the effect of adding solid potassium chloride to the system at equilibrium? Verify your answer experimentally.
- (vi) Why boiling tubes of same size are used in the experiment?

Kc Iron Thiocyanate Equation

Liyong Dong

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