13 Al -> "e + 26 Ma 48K -> = e + 40 G balance nuclear reaction equations aluminum-26 undergoes positron emission potassium-40 undergoes beta decay (electron) 238 () -> 4 He + 234 Th uranium-238 undergoes alpha decay electron configurations 152 25226 35236 451345 Write the electron configuration for C. Write the electron configuration for Cr. Write the shorthand notation for Bi. De] 65244145 110603 NOMENCLATURE [naming compounds, writing formulas] Write the formula for each of the following compounds: a. Sulfur hexafluoride SFS b. Lithium nitride Link c. Chromium (III) carbonate Cr2 ((0))3 d. Tin (II) chloride Sn Ce2 e. Ammonium acetate NHUCZHOR (OF CHOCOONHY) f. Mercury (I) chloride Hazelz g. Potassium bromate KBrog h. Hydrosulfuric acid H2 Some i. Chloric acid Sulfurous acid 42300 Name each of the following compounds. a Cuso, copper (II) sulfate aluminum flooride b. AIF. c. HI d. NO narosen monovale c. H₂Sc (40) hydroselen ic acid f. HNO, notric acid narous acrd g. HNO: NaHSO_a Soulden

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it takes for a 40.0 g sample to decay down to 1.25 g? 40 -> 20 -> 10 -> 5 -> 2.5 -> 1.25 ->

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

3 half lives 100-750-25-3 [2.5 mg]

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65,

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

63
$$\times .692 = 43.596 \rightarrow 49$$

65 $\times .309 = 20.02 \rightarrow 20.$
 $\frac{+}{63.62} \rightarrow 64$

K12 Chemistry A Laboratory Guide Answers

Marc le Maire, Raymond Chabaud, Guy Hervé

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Laboratory Guide to Biochemistry, Enzymology, and Protein Physical Chemistry Marc le Maire, Raymond Chabaud, Guy Hervé, 2012-12-06 The study of a single well chosen substance here aspartate transcarb amylase can provide an excellent basis for a laboratory course The student is introduced to a variety of scientific ideas and to many experi mental and interpretive techniques This enzyme is readily available is relatively stable has an extensive literature and its behavior has many facets substrate inhibition a large change in structure upon homo tropic activation by substrates allosteric stimulation by ATP allosteric inhibition by CTP synergistic with VTP positive cooperativity for sub strates negative cooperativity for CTP binding and dissociation and reassembly of subunits Cand R2 from the holoenzyme CI 5 In addition 3 6 to the known biochemical aspects of these properties the results ob tained here can be interpreted in the light of the high resolution X ray diffraction structures of the T and R forms the low angle X ray scattering results and the large number of mutants now available by recombinant DNA methods Future development of this course could also involve part of these methods as well as the carefully chosen experiments described here This approach resembles research more than the approaches one usually finds in biochemical laboratory courses A consistent develop ment of ideas about a single enzyme which shows so many facets in its behavior is sure to hold the interest of the student Moreover one explores a depth and The Complete Home Learning Sourcebook Rebecca reasons to move forward that are an essential part of research Rupp, 1998 Lists all the resources needed to create a balanced curriculum for homeschooling from preschool to high school Chemistry of Cement ,1962 Illinois Chemistry Teacher, 2007-09 Nuclear Science Abstracts, 1976-03 level **Resources in Education** ,1999-04 ENC Focus, 2001 The Science Teacher .2007 New Horizons in Mathematics and Science Education ,2001 Scientific and Technical Aerospace Reports, 1975 Lists citations with

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