Section A.

For each question there are four possible answers, A. B. C and D. Choose the one you consider to be correct.

- 4. Which equation correctly represents the enthalpy change of atomisation of iodine?
 - A Start Inc.
 - $B = 1_{local} \longrightarrow 21_{loc}$
 - $C = 1_{loc} - 21_{loc}$
 - 10 4 f ----- f
- What is the second electron affinity of oxygen?
 Use the data below to determine your answer.

	AH kErnel	
Exchalpy of formation of calcium oxide	-636	
Lattice embalpy of calcium oxide	-3401	
Second ionisation energy of calcium	+1100	
First ionisation energy of calcium	+550	
Enthalpy of atomination of calcium	+177	
Enthalpy of atcretiution of oxygen	+249	
First electron affinity of cryyges	-141	

- A: +790 kFunel.
- B = 790 kJ/mod.
- C +1 039 kJ/med
- D -1 039 kJ/mol
- 3 Why does the enthalpy change of neutralisation of oleic acid.

CH₂(CH₂)₂CH = CH₂(CH₂)₂ CO₂H, with approximation of differ from the theoretical value of -57.2 kJrant⁻¹?

- A Ofeic acid and ammonia do not react in the ratio 1:1.
- B Office acid is unsaturated therefore its enthalpy of neutralisation is more exorternic.
- C Energy is required to ionise ofeic acid and ammonia.
- D Ammonia is completely hydrolysed.

June 2013 Chemistry Paper 3

J Ma

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Unveiling the Magic of Words: A Overview of "June 2013 Chemistry Paper 3"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly aweinspiring. Enter the realm of "June 2013 Chemistry Paper 3," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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