

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

$$\begin{aligned}\text{Gas constant } R &= 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1} \\ &= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}\end{aligned}$$

$$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J} \quad 1 \text{ Faraday} = 96500 \text{ coulombs.}$$

$$\text{Avogadro's number} = 6.023 \times 10^{23}$$

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

June 2013 Chemistry 97paper 22 Question

Michael Hoelscher



June 2013 Chemistry 97paper 22 Question:

This book delves into June 2013 Chemistry 97paper 22 Question. June 2013 Chemistry 97paper 22 Question is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into June 2013 Chemistry 97paper 22 Question, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to June 2013 Chemistry 97paper 22 Question
- Chapter 2: Essential Elements of June 2013 Chemistry 97paper 22 Question
- Chapter 3: June 2013 Chemistry 97paper 22 Question in Everyday Life
- Chapter 4: June 2013 Chemistry 97paper 22 Question in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of June 2013 Chemistry 97paper 22 Question. The first chapter will explore what June 2013 Chemistry 97paper 22 Question is, why June 2013 Chemistry 97paper 22 Question is vital, and how to effectively learn about June 2013 Chemistry 97paper 22 Question.

3. In chapter 2, the author will delve into the foundational concepts of June 2013 Chemistry 97paper 22 Question. This chapter will elucidate the essential principles that need to be understood to grasp June 2013 Chemistry 97paper 22 Question in its entirety.

4. In chapter 3, the author will examine the practical applications of June 2013 Chemistry 97paper 22 Question in daily life. The third chapter will showcase real-world examples of how June 2013 Chemistry 97paper 22 Question can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of June 2013 Chemistry 97paper 22 Question in specific contexts. This chapter will explore how June 2013 Chemistry 97paper 22 Question is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about June 2013 Chemistry 97paper 22 Question. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of June 2013 Chemistry 97paper 22 Question.

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