

সেট-০১

জুনিয়র ফুল সার্টিফিকেট পরীক্ষা

বাংলা (সৃজনশীল + নির্দিষ্ট + বহুনির্বাচনি অঙ্গীকৃত)

বিষয় কোড : 1 0 1

সময়—৩ ঘণ্টা

পূর্ণমান—৪০+৩০+৩০=১০০

সৃজনশীল প্রশ্ন

মান—৪০

দ্রষ্টব্য : জ্ঞান পাশের সংখ্যা প্রশ্নের পূর্ণমান জ্ঞাপক। প্রদত্ত উদ্দেশ্যবাক্যে মনোযোগসহকারে পড় এবং সংশ্লিষ্ট প্রশ্নের উত্তর দাও। যে কোনো চারটি প্রশ্নের উত্তর দিতে হবে।]

পদ্যাংশ

(যে কোনো দুটি প্রশ্নের উত্তর দাও)

মান—১০×২=২০

নম্বর

- ১। আনোয়ার সাহেব অনেক পরিচয় করে সত্যতার সাথে ব্যবসা করে প্রতিষ্ঠিত হয়েছেন। কোনোদিনও কারো ক্ষতি করেননি। পরিচয় আর সত্যতাই ছিল তাঁর প্রেক্ষাপট। পক্ষান্তরে রজব আলী সুদের কারবার করে ধনাঢ্য ব্যক্তিতে পরিণত হয়েছেন। মানুষকে বিপদে ফেলে চড়া সুদে টাকা ধার দেয়াই তার প্রধান লক্ষ্য। শেষ ব্যসে এসে তার মনে কোনো সুখ নেই। তার ভেতরে শুধুই অতৃপ্তি আর হাহাকার।
- (ক) 'সুখী মানুষ' নাটিকায় মোড়লের কথিত রোগের নাম কী?
- (খ) সুখকে বড় কঠিন জিনিস কহা হয়েছে কেন?
- (গ) উদ্ভীপকের রজব আলীর সাথে 'সুখী মানুষ' নাটিকার সাদৃশ্যপূর্ণ চরিত্রটি ব্যাখ্যা কর।
- (ঘ) উদ্ভীপকের আনোয়ার সাহেব এবং 'সুখী মানুষ' নাটিকার কাঠুরে মেন এক অভিন্ন সত্তা—যুক্তিসহ বুঝিয়ে লেখ।

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- ২। নির্দিষ্ট অবসান ঐ পুরাতন বর্ষ হও গত
বশু হও শত্রু হও যেখানে যে রহ
অম্মা করিও আজিকার মতো
পুরাতন বর্ষের সাথে পুরাতন অপরাধ মতো।
- (ক) বাংলা নববর্ষের প্রধান অনুষ্ঠানের নাম কী?
- (খ) 'বাংলা নববর্ষ আমাদের প্রধান জাতীয় উৎসব'—কেন?
- (গ) উদ্ভীপকের ভাবার্থ 'বাংলা নববর্ষ' প্রবন্ধের কোন নিকটি নির্দেশ করে? ব্যাখ্যা কর।
- (ঘ) উদ্ভীপকটি 'বাংলা নববর্ষ' প্রবন্ধের সমগ্র ভাব ধারণ করে না।—বিশ্লেষণ কর।

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- ৩। করিম ও রহিম দুই ঘনিষ্ঠ বন্ধু। টাকা উত্তোলনের জন্য ব্যাংকে গেলেন। কাউটারে একলক্ষ টাকার চেক মিলে ক্যাশিয়ার সাহেব ভুলে দশ হাজার টাকা বেশি দিলেন। দুই বন্ধু বাসায় এসে দেখলেন দশ হাজার টাকা বেশী। করিম সাহেব বললেন অতিরিক্ত টাকা আমরা ফেরত দিবো না। নিজেরা খরচ করবো। কিন্তু রহিম সাহেব বললেন বাকি টাকা ফেরত দিবো। কেননা এ অতিরিক্ত টাকার জন্য ক্যাশিয়ার সাহেবের চাকরি চলে যেতে পারে। পরে দুই বন্ধু মিলে টাকাগুলি ফেরত দিলেন।
- (ক) 'পড়ে পাওয়া' গল্প কে সকলের সংশয় দূর করে দিল?
- (খ) অধরপুরের কাপালিরা সর্বদা হুমুসে কীভাবে?
- (গ) উদ্ভীপকের করিম ও রহিম 'পড়ে পাওয়া' গল্পের কাহিনীর প্রতিনিধিত্ব করে? ব্যাখ্যা কর।
- (ঘ) উদ্ভীপকের ক্যাশিয়ার 'পড়ে পাওয়া' গল্পের কাপালি একই সূত্রে গাঁথা—বিশ্লেষণ কর।

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কবিতাংশ

(যে কোনো দুটি প্রশ্নের উত্তর দাও)

মান—১০×২=২০

- ৪। হোসেন একজন বর্ণাচারী। অন্যের জমি চাষাবাদের মাধ্যমে তার সংসার চলে। একমাত্র মেয়ে রহিমাকে দেখাপড়া শেখানোর জন্য ছুপে পাঠায় কিন্তু বিত্তবান রহমত আলী বিষয়টা ভালোভাবে নেয় না। ঘড়ঘড় করে রহমত তাকে ভিটে-মাটি ছাড়ার করে। হোসেন নিরুপায় হয়ে দূর গ্রামে চলে যায়। রহমত আলীর বিশ্ববিদ্যালয় পড়মা ছেলে অমিত পিতার কর্মকাণ্ডে বিরক্ত ও লজ্জিত হয় এবং পিতাকে এ ধরনের অমানবিক আচরণ থেকে বিরক্ত করে হোসেনকে তার ভিত্তীয় ফেরার ব্যবস্থা করে।
- (ক) 'দুই বিঘা জমি' কবিতায় উপেন কতদিন পর ভিটে-মাটি ছেড়ে পথে বের হল?
- (খ) 'তুমি মহারাজ সাধু হলে আর'—একথা বলার কারণ কী?
- (গ) উদ্ভীপকের রহমত আলীর সাথে 'দুই বিঘা জমি' কবিতার সাদৃশ্যপূর্ণ চরিত্রটি ব্যাখ্যা কর।
- (ঘ) 'দুই বিঘা জমি' কবিতার উপেন যদি উদ্ভীপকের হোসেনের অবস্থায় পড়তেন তাহলে উপেনকে ভিটে-মাটি ছাড়তে হতো না।—মন্তব্যটি বিশ্লেষণ কর।

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Jsc Exam 2014 All Subject Question Dhaka Board

Michael Brown



Jsc Exam 2014 All Subject Question Dhaka Board:

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<https://staging.conocer.cide.edu/data/browse/Documents/grumpy%20old%20women.pdf>

Table of Contents Jsc Exam 2014 All Subject Question Dhaka Board

1. Understanding the eBook Jsc Exam 2014 All Subject Question Dhaka Board
 - The Rise of Digital Reading Jsc Exam 2014 All Subject Question Dhaka Board
 - Advantages of eBooks Over Traditional Books
2. Identifying Jsc Exam 2014 All Subject Question Dhaka Board
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jsc Exam 2014 All Subject Question Dhaka Board
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jsc Exam 2014 All Subject Question Dhaka Board
 - Personalized Recommendations

- Jsc Exam 2014 All Subject Question Dhaka Board User Reviews and Ratings
- Jsc Exam 2014 All Subject Question Dhaka Board and Bestseller Lists
- 5. Accessing Jsc Exam 2014 All Subject Question Dhaka Board Free and Paid eBooks
 - Jsc Exam 2014 All Subject Question Dhaka Board Public Domain eBooks
 - Jsc Exam 2014 All Subject Question Dhaka Board eBook Subscription Services
 - Jsc Exam 2014 All Subject Question Dhaka Board Budget-Friendly Options
- 6. Navigating Jsc Exam 2014 All Subject Question Dhaka Board eBook Formats
 - ePub, PDF, MOBI, and More
 - Jsc Exam 2014 All Subject Question Dhaka Board Compatibility with Devices
 - Jsc Exam 2014 All Subject Question Dhaka Board Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jsc Exam 2014 All Subject Question Dhaka Board
 - Highlighting and Note-Taking Jsc Exam 2014 All Subject Question Dhaka Board
 - Interactive Elements Jsc Exam 2014 All Subject Question Dhaka Board
- 8. Staying Engaged with Jsc Exam 2014 All Subject Question Dhaka Board
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jsc Exam 2014 All Subject Question Dhaka Board
- 9. Balancing eBooks and Physical Books Jsc Exam 2014 All Subject Question Dhaka Board
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jsc Exam 2014 All Subject Question Dhaka Board
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Jsc Exam 2014 All Subject Question Dhaka Board
 - Setting Reading Goals Jsc Exam 2014 All Subject Question Dhaka Board
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jsc Exam 2014 All Subject Question Dhaka Board
 - Fact-Checking eBook Content of Jsc Exam 2014 All Subject Question Dhaka Board

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655

furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1
 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web Jul 19 2023 introduction titration is an analytical quantitative technique used to determine the concentration of a solute a pH titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH

reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ $4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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