

Name _____ Date _____ Period _____

VIDEO GUIDE: NOVA – “Hunting the Elements”



A small, stylized periodic table of elements, showing the layout of the periodic table with elements grouped into periods and groups.

- 1) How many different elements are there? _____
- 2) What is the symbol for gold? _____
- 3) What is gold's atomic number? _____
- 4) What are some unique properties of gold? _____
- 5) What process is used to separate gold atoms from crushed rock? _____
- 6) How much gold is harvested per ton of earth from the underground mine? _____

7) Gold does not share its electrons easily. Therefore, it is not very chemically reactive. What are two other metals that, like gold, don't react with other elements? _____

8) What is the worth (\$\$) of one 80-pound bar of gold? _____

9) Two rows above gold is the metal: _____; its symbol is: _____

10) In what objects / industries is copper used? _____

11) What are some unique properties of copper? _____

12) What is the symbol for tin? _____ What is its atomic number? _____

13) When copper and tin are mixed together, the alloy _____ is formed.

14) Why isn't copper alone used to make bells? _____

15) It is possible that the crack in the famous Liberty Bell was caused by too much _____

16) Metals form crystals, which means that they have an _____ of atoms.

17) Atoms with 1 proton are the element: _____; atoms with 2 protons are _____.

18) The number of protons in an atom is known as that element's _____.

19) The chemical symbol of each element is based on the element's name in which language? _____

20) More than 70% of the elements on the periodic table are _____.

21) Why are the noble gases called "noble"? _____

22) Which subatomic particles determine an element's reactivity? (circle correct answer)

PROTONS

NEUTRONS

ELECTRONS

23) How many electrons are in the first shell of electrons? _____ The second? _____

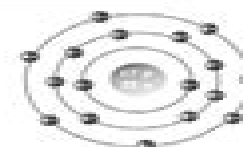
24) How many electrons do the noble gases have in their outermost shell? _____

25) How many electrons does chlorine have in its outermost shell? _____

26) What will an atom of chlorine do to complete its outer shell? _____

27) The alkali metals, such as sodium, have _____ electron in their outer shell.

28) What do the alkali metals do with that one electron? _____



Hunting The Elements Nova Worksheet Answers

Sabine Zange



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