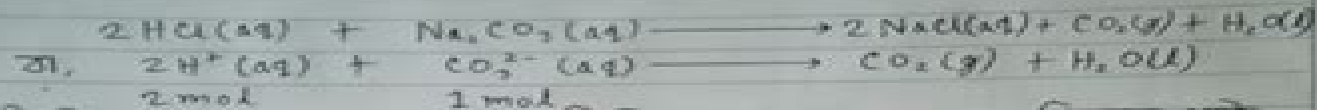


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্যম দ্বারা সংকলিত নমুনা উদ্যম (HCl/উদ্যম)	Expt No.	01
চলমান দিন	Date :	

মূলনীতি: Na_2CO_3 ও HCl এর প্রকৃত বিক্রিয়ার সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃত বিক্রিয়া কোথায় প্রকৃত বিক্রিয়ার দিকে, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যালি) থেকে দেয়া।

প্রয়োজনীয় সামগ্র্যনিক বস্তু:

- ১) 0.1M Na_2CO_3 দ্রব
- ২) 0.1M HCl দ্রব
- ৩) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- ১) কনিক্যাল ফ্লাস্ক
- ২) ব্যালো
- ৩) পিপেট
- ৪) ক্যাম্পাস মেসার

কার্যের ধারা:

- ১) প্রস্তুত করা 0.1M HCl দ্রব (প্রকৃত দ্রব) অথবা নমুনা HCl দ্বারা ব্যালোটি বর্ণ করে এবং মেসারের মাধ্যমে ক্যাম্পাস দ্বারা পরিমাপ করে আটকিয়ে নিই।
- ২) প্রস্তুত করা 0.1M Na_2CO_3 দ্রব (প্রকৃত দ্রব) এর আনুমানিক ফ্রাক্ট থেকে পিপেট দ্বারা 10ml দ্রব কনিক্যাল ফ্লাস্কে নিই। কনিক্যাল ফ্লাস্কে দ্রব 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, তখন দ্রবের বর্ণ হলুদ হয়।
- ৩) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা কাগজ টুকরা রাখি, ব্যালোটি HCl এর দ্রবের দিকে নিয়ে ঘোঁরা দিতে '১ম পাঠ' - এ ফেরত করে। চিত্রমাতে তখন মাঝে ব্যালোটির দৈর্ঘ্য তখন ঘোঁরা ব্যালোটির HCl দ্রব ঘোঁরা ফোঁটা কনিক্যাল ফ্লাস্কে যোগ করে

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Siddappa N.Byrareddy



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