BACKS TOWNSON, S NIV 30 5 x 1 M = 0.5479 M MEAN, IN I BUTHE WANTED

Stee we, w - 20 m HICH HIS SEMINE WILL M = 36.5 g/medHarvis minor, $\nabla = 11$. Herefore represents $N \ll \gamma$

্রা কুলাকে বালিও X, লাকাভিতে 10 mg এক কেছে কম পরিমাণ দ্বাৰ $_{\rm B}$ $_{
m V}$ -symblectic $_{
m 0.5}$ g $_{
m 200}$ $\left(\frac{M}{V}\right)$ 1 K.1 Weyer Stein mins and strains : কুলা X. পৃথাতিটি হলে সেমি হাইকো আন্দানাইটিকাল পৃথাতি ও ত্রতার হলো মাতো আনগাইটিকাল ককতি। উত্তর কতিত্র মুক্তিকাল ককটি পরিবেশ কক্ষেত্র চিক্ত ত্ব প্ৰতিয়োজন কৰিছিল কৰিছেল আৰুৰ। নিয়ে তা বিয়েছণ কৰা

ক্রমেইরের পাবতি রামায়নিক বুলগত বিরেছদের একটি পাবতি। ক্রমেইরের পাবতি বামায়নিক বুলগত বিরেছদের একটি পাবতি। ুত বেছামাটার বসায়নে মহা পরিমাশ রামায়নিক উপাদনে স্নায়করণ ও ক্রিলেকে ডিমেবল বাবস্থত হয়। এ পশ্বতিতে সুব তার পরিমাণ (0.05 ত্রত (i.2g তর্তুন বা 2-4mil. তরণ প্রবা) রাম্বর্জনিক শদার্থ নিয়ে ্তিত করা হয়। কলে পতিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আদে। সেমি মাইজো পাবাতিহত 111,58 बार शरिकार्ड তার অনুসামাইত ব্যবহার করা হয় কলে পরিবেশ দূষণ ঘটে না। CH.CSNH; + H₂O ---- CH₂CONH₂ + H₂S

_{লিকা} খোনো কোনো কেতে কম ভূকির রাসায়নিক দুবা বাবস্তুত বয়। কলে পরিবেশ দুঘণের সম্ভাবনা কমে আলে। তাই সেমি-মাইজ্রে পৃস্পতি কটোৱেল কলেবার ।

ELLES



- ত, তাইভার প্রথক কীপু
- সেমি-মাইক্রো পশ্বতি পরিবেশ বান্ধর কেন?
- ্ ত্রণটির জন্য প্রযোজ্য ম্যাজার্ড প্রতীকটি অচকন করে। এবং यहारता स्ट्रहा ।
- ্ উন্নিপকের আলোকে দ্রবদ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে IM ঘনমাত্রার প্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪

৩ নং প্রয়োর উত্তর

🌌 বিশেষণীয় নিস্থির বীমের উপর রাইভার স্থাপন করলে বীমের প্রতি ^{মান্তির} জন্য যে ভর পাওয়া যায়, তাকে রাইভার ধুবক বলে।

🔯 সেমিনইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। ্ট বিচেম্পির রসায়নে ভয় পরিমাপ রাসায়নিক উপাদান সনাজকরণ ও র্থনিন্দত বিশ্লেষ্ট্র ব্যবস্থৃত হয়। এ পন্ধতিতে দুব অয় পরিমাণ ^{(0,05} সেকে 0.2g কঠিন বা 2-4ml, তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীয়া করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে। ^{ছাবন্ত}, কোনো কোনো কেন্তে কম ঝুঁকির রাসায়নিক মধ্য ব্যবহৃত হয়। ^{মাস পরিবেশ} দূরণের সম্ভাবনা কমে আসে। তাই সেমি-মাইরেন পশ্বতি পরিবেশ সংস্থাব ।

ট্র ইমিশরের প্রবর্গটি হলো সালফিউরিক এপিড (H₂SO₄) এর 10 M ইল। H₅SO₄ একটি জাত্রক ও কয়কারী পদার্থ। H₂SO₄ এর 10 M ইল হাৰা পাচ H₂SO₄ স্তবণ। গাঢ় H₂SO₄ পদাৰ্থকে ক্ষ্যাকাৰ্ত্তী আজাৰ্ত তিত্বৈ অনুসূত্র করা হয়। গাড় H2SO4 এর জন্য করকারী হাজার হটার নিয়ে অন্তব্দ করে ব্যাখ্যা করা হলো:

weeds (Corresive)
ক্ষরবাধী রাম্মান্ত্রিক উপাধান বলতে সে সকল রাম্মান্ত্রিক প্রবাধে বুরার যার সংস্পার্ক তিক ও জনানা পালার্ক্তর মানাক্রক ক্ষতি সাধিত বহা। এতে তুক পুচে
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ট্রালাকের মুনশাটি হতের 1604 H₅SO, প্রদান এবং এর বেকে 1M ঘনমান্ত্ৰত প্ৰথম প্ৰায়ত কৰাত মাল ১৯ ৫ ৩৪ কাড্ৰয় বাংলাৰ কৰাত

১৭ কাচয়ন্ত্ৰটি হলো 100 ml. আহাতনমিতিক ভ্ৰাণক ও এই টি হলো পিলেট। যেমেতু আয়তনমিতিক য়াংশেরর আয়তন 100 mt., তাই আমানের 100 mL IM খনমাত্রার H₋SO, প্রবদ তৈরি করতে হবে : দ্রবংগর অধুকরণ সূত্র যতে আমরা লাগি,

$$V_1S_1 = V_2S_2$$

 $V_1 = \frac{V_2S_2}{S_1}$
 $= \left(\frac{100 \times 1}{10}\right) \text{ mL}$
 $= 10 \text{ mL}$

আৰ্থমিক আন্তচন, ৮, – ৫ $484121, S_1 = 10M$ CON SCHOOL V2 = 100 mL weight, $S_2 = 1M$.

অক্তর্যর, 10mL H₃SO, স্তব্যের সচেদ পানি যোগ করতে মনে (100-10) mL = 90 mL

আবার, শিপেটের স্মায়ায়ে 10 mL প্রদান খনমাত্রার প্রবণ নিচে তার সাবে আয়তনমিতিক ফ্লান্ডে 90 mL পানি যোগ করলে আমালের কাজিনত খনমাত্রার দ্রবণ পাওয়া যাবে ৷

EH CHA







538 - A.

Per Line Day II with the Art

- ক, আন্টিঅক্সিডেট কাকে বলেং
- थ. माथा धूरक ८, धर यान कथाना अभीय रह गा–साथा PERCENT I
- A ভিত্তর মন্ত্রটিকে উত্তর করার কৌশল বর্ণনা করা।
- R ত C চিছের সাথে সম্পর্কিত রাসায়নিক প্রবাসির ব্যবহারকালে দ্বাস্থ্য কৃতি স্তাদে করণীয় উপায় সম্পর্কে विद्रायम् करता ।

ह भर् कार्यात किन्द्र

আটিঅরিডেউ এক ধরনের রাসারনিক বা কৃত্রিম খাদা শবেকক য়া মূলত থাৰা উপাদানসমূহকে জাৱণ প্ৰতিষ্কায় তাৱন থেকে ৱকা করে থাবার সংরক্ষণ করে।

সাম্যাতক K, এর মান কথনো অধীম বা পূন্য হতে পারে মা। উত্তিটি স্থাখ্যা করার জন্য নিচের উতমুখী সমীকরণটি বিবেচনা করি— $A + B \rightleftharpoons C + D$

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