SCHEME OF WORK 2007

CHEMISTRY

FORM FIVE

FIRST TERM -03.01.2007 to 09.03.2007 (10 weeks) 19.03.2007 to 25.05.2007 (10 weeks)

THEME LEARNING AREA ENTERACTION REPORTS CHEMOCALS 1 RATE OF REACTION

| No. of week/ Weeks | Learning Oligedbyes | Suggested Learning Activities | Learning Outcomes | Thinking Skills | Scientific Skills | Scientific Atthudes And Noble Values | Remarks/ 'Woaltuilary |
|--------------------------|-----------------------------|---|--|---|--|--|--|
| | Prusiyoing rate of reaction | Discours: (a) the meaning of rate of reaction. (b) some examples of first reactions. (c) some examples of first reactions. (d) some examples of slow reactions. Discours to identify observable changes to reactions. Discours to identify observable in order to determine the rate of reaction. Carry out an activity involving a reaction between sinc and acid, and plot a graph to determine average rate of reaction at any given time. Carry out problem solving activities involving a resolving activities involving rates of reaction. | A student is able to: state what rate of reaction is. lidentity observable changes to reactants or products for determining rate of reaction. determine average rate of reaction, determine the rate of reaction, determine the rate of reaction at any given time from a graph. solve numerical problems average rate of reaction, tolve numerical problems involving average rate of reaction. solve mumerical problems for a solve involving rate of reaction. | generating ideas relating making inpotheral realising conclusion protoken solving | Observing Inspothershing Communication S Interpreting data | Being hone st and accounts in ne-conding and uslidating data. Being cooperative Being systematic | reactant brotain balas balas product- masil findsin balas rate of reaction balas sobservable change persubahan yang dayar diperhatikan |

1

| No. of week? Weeks | Learning Objectives | Suggested Learning Activities | Learning Outcomes | Thinking Skills | Scientific Skills | Scientific Attitudes And Noble Values | Pernanks/ Viscabulary |
|--------------------------|---|--|--|--|---|--|--|
| 2 | 1.2 Synthesising declars affecting the rate of reaction | Discouse possible factors affecting the rate of reaction. Design and carry-out activities to investigate factors affecting the rate of reactions, i.e. size of reactions, i.e. size of reactions, i.e. size of reactions and catalyst. Some suggested reactions: (a) a reaction between calcium cartionate. CaCO., and hydrochioris acid. HCI, (b) a reaction between sodium cartionate. Na.5.O., and sulphusic acid. HCS.O., (c) decomposition of hydrogen personice, HCO., in the presence of a catalyst. | A student is able to: design experiment to investigate factors affecting the rate of reaction. give examples of reactions that are affected by size of reactions temperature and catalyst. | generation g ideas relating enabling inpothersi si sanatyring contribution of the cont | predicting predicting date of the second seco | Being honest in necording and validating data Being systematic being cooperative being confident and independent | cuttalydt - in angliën de comporation- penguraian |
| | | Uses computer attimulations to innestigate how the movement and collision of granticles in a reaction are affected by temperature, size of reactant, pressure, concentration and catalyst. Collect and interpret data to explain factors affecting the rate of reaction in the fallowing. | englain how each Sector affects the rate of reaction. describe how Sectors affecting the rate of | | | | |

Federal Scheme Of Work For Sschemistry

K Morrison

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