

# HYDROCARBONS



## Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

### a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons :

- Alkane : open chain ( $C_nH_{2n+2}$ )
- Cycloalkane : closed chain ( $C_nH_{2n}$ )

### b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond ( $C_nH_{2n}$ )
- Alkyne : triple bond ( $C_nH_{2n-2}$ )
- Alkadiene : two double bonds ( $C_nH_{2n-2}$ )



## Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of alkanes from  $C_1$ - $C_{10}$  :

| Formula     | Name    | Formula        | Name    |
|-------------|---------|----------------|---------|
| $CH_4$      | methane | $C_6H_{14}$    | hexane  |
| $C_2H_6$    | ethane  | $C_7H_{16}$    | heptane |
| $C_3H_8$    | propane | $C_8H_{18}$    | octane  |
| $C_4H_{10}$ | butane  | $C_9H_{20}$    | nonane  |
| $C_5H_{12}$ | pentane | $C_{10}H_{22}$ | decane  |

Alkene names :

$C_2H_4$  = ethylene

$C_3H_6$  = propene

# Hydrocarbons Study Guide

**WJ Hussar**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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## **Table of Contents Hydrocarbons Study Guide**

1. Understanding the eBook Hydrocarbons Study Guide
  - The Rise of Digital Reading Hydrocarbons Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Hydrocarbons Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hydrocarbons Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hydrocarbons Study Guide
  - Personalized Recommendations
  - Hydrocarbons Study Guide User Reviews and Ratings
  - Hydrocarbons Study Guide and Bestseller Lists
5. Accessing Hydrocarbons Study Guide Free and Paid eBooks
  - Hydrocarbons Study Guide Public Domain eBooks
  - Hydrocarbons Study Guide eBook Subscription Services
  - Hydrocarbons Study Guide Budget-Friendly Options

6. Navigating Hydrocarbons Study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Hydrocarbons Study Guide Compatibility with Devices
  - Hydrocarbons Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hydrocarbons Study Guide
  - Highlighting and Note-Taking Hydrocarbons Study Guide
  - Interactive Elements Hydrocarbons Study Guide
8. Staying Engaged with Hydrocarbons Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hydrocarbons Study Guide
9. Balancing eBooks and Physical Books Hydrocarbons Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hydrocarbons Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Hydrocarbons Study Guide
  - Setting Reading Goals Hydrocarbons Study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hydrocarbons Study Guide
  - Fact-Checking eBook Content of Hydrocarbons Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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