

Guess who

Are they male or female?.....

Which Country are they from?.....

What is their nationality?.....

Which city are they from?.....

How old are they?.....

What's their job?.....

Trivia:.....

Guess who

Are they male or female?.....

Which Country are they from?.....

What is their nationality?.....

Which city are they from?.....

How old are they?.....

What's their job?.....

Trivia:.....

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Trivia:.....

How To Become A Celebrity English Edition

Miguel de Cervantes Saavedra



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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{CrO}_4^{2-} \text{aq} \rightarrow \text{Ag}_2\text{CrO}_4 \text{s}$

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web the six no 3 aq ions and the six na aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} + 2\text{aq} + 2\text{po}_4^{3-} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O l}$ the spectator ions in KOH aq and $\text{HNO}_3 \text{ aq}$ are $\text{K}^+ \text{ aq}$ and $\text{NO}_3^- \text{ aq}$ how to identify and list the

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