

Chemistry 1211 Final Exam Guide

Chapter 1:

- **matter:** anything that has mass
- **macroscopic:** can be seen, with the naked eye
- **microscopic:** can't see with the naked eye
- **element (atom/molecule):** cannot be broken down into anything simpler
- **Compound:** a pure substance that contains two or more elements combined together in a constant ratio
- **substance:** matter with a definite composition
- **Mixtures:** composed of two or more substances
- **Chemical compound:** Matter that is composed of two or more kinds of atoms, chemically combined in definite proportions
- **Molecules:** The smallest unit of a compound that retains the composition and properties of that compound
- **Pure substance:** A substance that contains only one type of substance or compound and has fixed chemical composition
- **States of matter:**
 - o **Gas:** a gaseous material with a defined shape, atoms or molecules are widely spaced together; molecules are moving fast; high energy; good mix
 - o **Liquid:** atoms and molecules are close to the container, atoms or molecules are close together but they move good mix
 - o **Solid:** has a fixed shape or volume; atoms or molecules are tightly spaced apart and are moving rapidly from one position to another; particles don't vibrate; interact with each other and move slowly from one position to another
- **Gas:** ☐ liquid: from a certain height
- **Liquid:** ☐ solid: from a certain height
- **Liquid:** ☐ gas: from a certain height
- **Physical change:** a change in physical property
- **Chemical change:** a change in chemical property
- **Mixture:** a substance made up of two or more elements or compounds, but not necessarily chemically
 - o **Heterogeneous mixture:** composed of two or more different substances
 - o **Homogeneous mixture:** composed of two or more different substances

Georgia Standardized Final Exam Chemistry 1211

Jianjun Gao



Georgia Standardized Final Exam Chemistry 1211:

Reviewing **Georgia Standardized Final Exam Chemistry 1211**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Georgia Standardized Final Exam Chemistry 1211**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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