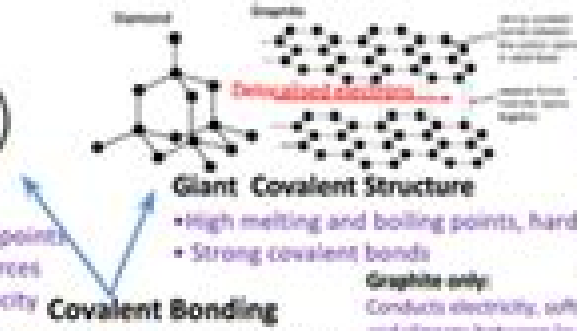
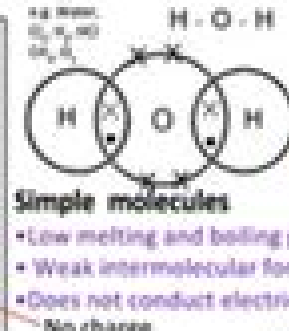




Periodic Table of the Elements

Legend:

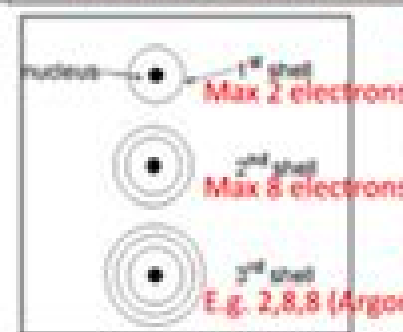
- metals
- non-metals
- metalloids
- transition metals
- alkali metals
- alkaline earth metals
- transition metals
- lanthanoids
- actinoids

An atom:
Number of Protons = Number of Electrons
Mass number = Protons + Neutrons

C2 Chemistry

Periodic table arranges elements by
ATOMIC NUMBER (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
E.g. Lithium 2,1 is a group 1 element.

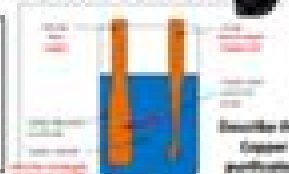


Protons and neutrons both have a mass of 1

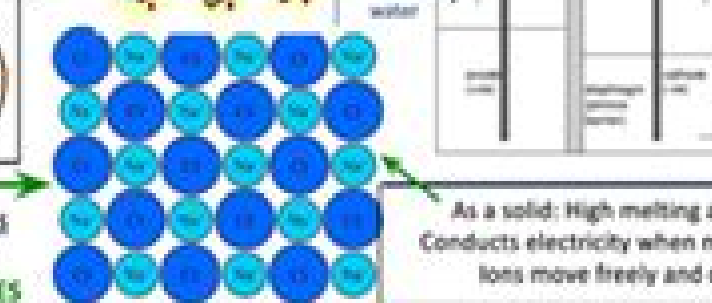
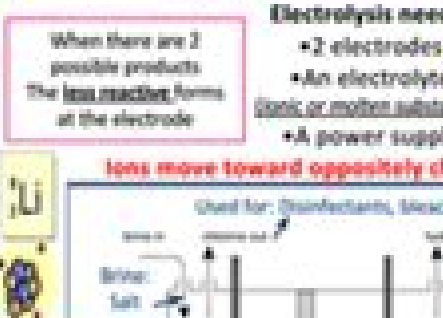
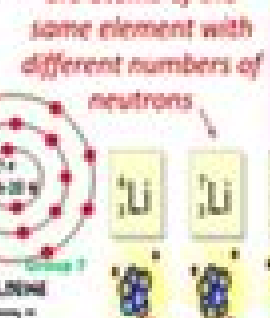
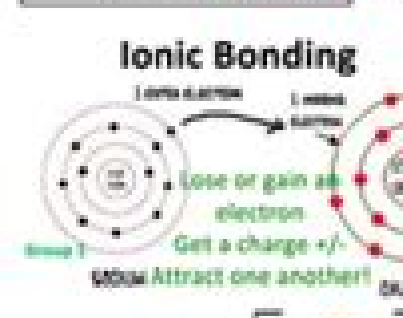
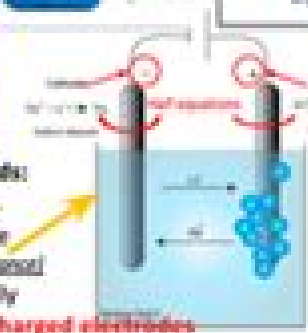
ISOTOPES
are atoms of the same element with different numbers of neutrons

OIL RIG

Oxidation is loss (of electrons)
Reduction is gain (of electrons)



Electrolysis:
Splitting up a substance using electricity



NANO-TECHNOLOGY
Means REALLY REALLY REALLY small!!!!

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graduation student s year of entry into grade 9 2010 2011 classes of 2014 2015 2012 and on class of 2016 and on n j
department of education required assessments nj hspa 11th grade beginning in 2014 2015

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