Calculate the enthalpy of combustion of ethane, C₂H₆, at 25 °C and one atmosphere pressure.

$$C_2H_6(g) + 3.5O_2(g) \rightarrow 2CO_2(g) + 3H_2O(l)$$

Compound	$\Delta H_{\rm f}^{\rm o}$, kJ·mol·1
$C_2H_6(g)$	-84.7
$H_2O(l)$	-285.8
$CO_2(g)$	-393.5

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J.F. Griffiths

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web start by identifying the main characters in the story and their initial goals or motivations 02 introduce a sudden and unexpected plot twist that alters the course of the story 03