

Variability and Sampling

Data is everywhere. In its rawest form, it's in our science classes, it shows up in economics, statistics and anywhere else that variability occurs. Variability is simply the property of outcomes being different. The tools of statistics are designed to explain this variability. There are many types of variability. It is good to understand these various tools in order to understand their strengths and weaknesses.

Example 1 The following types of variability can change outcomes of a data set. For each, give an example that may hold.

- (a) **Observational or Measurement Variability:** Variability that is introduced due to either our measuring instruments not being precise enough or differences in how individual people read the measurement.

2 students measure time for a ball to roll down a ramp
→ different reaction times

- (b) **Human Variability or Human-Induced Variability:** Variability that occurs due to the fact that members of populations are simply different.

2 people work out same amount of time
→ different people use different amounts of weight

- (c) **Random Variability:** This type of variability is the result of noise and error. It can be because our tools assigned our populations or samples to two or more treatment groups and then observing the variability between the groups.

2 groups taking a test, one group allowed to sleep 8 hours, other group only 4 hours → different tests

- (d) **Sample Variability:** This is the type of variability that occurs when we take multiple samples. There is population variability. There is going to still be difference due to the variability of the samples or errors.

Multiple samples of people surveyed to determine political candidate preference

Remember, through all of our work in this book, we are really trying to explain the variability of data within either a population or a sample and then using this to determine if the variability was the introduced by one of the factors above or the variability of the others.

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1/267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1/273 and concluded that the pressure would approach zero at 1/273 degrees c

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web the volume of 1.00 mol of any gas at stp standard temperature 273.15 K and pressure 1 atm is measured to be 22.414 L we can substitute 101.325 kPa for pressure 22.414 L for volume and 273.15 K for temperature into the ideal gas equation and solve for R

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273.15 K and 1 atm 101.325 kPa 1 atm at stp one mole of an ideal gas has a volume of about 22.4 L this is referred to as the standard molar volume figure 9.18

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