

QUESTION 1

- 1.1
- Fires
 - Injury because of protruding nails, splinters, etc.
 - People tripping over loose objects on floors
 - Objects falling from above
 - People bumping against poorly piled or badly placed materials
 - People slipping on greasy, wet or dirty floors
- (Any 3 × 1) (3)
- 1.2 An accident is an unplanned event that can result in an injury✓, the loss of life or damage to property or goods. ✓✓ (3)
- 1.3
- 1.3.1 They are switches or contacts included in a circuit to ensure safety by causing a circuit to be opened in the event of an incorrect operation.
- 1.3.2 It is a switch, which in the open position, satisfies the isolating requirements specified for a disconnecter.
- (Any 2 × 2) (4)
- [10]**

QUESTION 2

- 2.1
- Open flames
 - Heated surfaces
 - Electrical faults
 - Friction
 - Cigarettes and matches
 - Lightning
 - Static electricity
 - Chemical reactions
- (Any 5 × 1) (5)
- 2.2 A power tool requires additional power over and above manual labour. (1 × 2) (2)
- 2.3
- Clean power tools with a cloth after use.
 - Store tools in a clean, dry place.
 - Electric cords should be checked before a power tool is used.
 - Ensure that the battery is packed according to manufacturer's specification.
 - Blades and other cutting accessories should be sharp.
 - Tighten components that might have been shaken loose during operation.
 - Blow all air vents clean.
 - Lubricate according to the instructions in the user manual.
- (Any 3 × 1) (3)
- [10]**

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Lingjun Ying



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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction e.g. particularly nitration of methyl benzoate to give methyl m-nitrobenzoate thus determine the weight, melting point and percentage yield of the pure methyl m-nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene. Bromine substituted on an aromatic ring is an ortho/para directing group. It will direct an incoming electrophile to the ortho and para positions. It is also deactivating, meaning that its reactions will be slower than the identical reactions of benzene.

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule. Nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid. This mixture is useful to obtain the active nitronium ion.

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1 go to the restroom and wash the affected area with lots of soap and water. The soap in the lab is not suitable for this purpose. 2 return to the lab and apply mineral oil to the affected area. 3 the summary to this warning is not to touch any part of your body while performing this experiment. If you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another. Nitric acid, the weaker of the two acids, will serve as base to form nitronium ion. Bromobenzene will attack the nitronium ion by nucleophilic attack.

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show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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nitrobromobenzene recrystallization was used to purify 4 nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group