Balancing Chemical Reactions - Practice

Balance the following reactions and identify what type of reactions they are.

1)
$$1 N_2 + 3 H_2 \rightarrow 2 NH_3$$

Type of Reaction? ____ Synthesis



Type of Reaction? Double Replacement

3)
$$1 P_4 + 3 O_2 \rightarrow 2 P_2O_3$$

Type of Reaction? Synthesis

4) $2 \text{ RbNO}_3 + 1 \text{ BeF}_2 \rightarrow 1 \text{ Be(NO}_3)_2 + 2 \text{ RbF}_3$

Type of Reaction? ____ Double Replacement

5) $2 \text{ AgNO}_3 + 1 \text{ Cu} \rightarrow 1 \text{ Cu(NO}_3)_2 + 2 \text{ Ag}$

Type of Reaction? Single Replacement

6) $2 \text{ H}_2\text{O}_2 \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ O}_2$

Type of Reaction? Decomposition

7) 2 HCN + 1 CuSO₄ \rightarrow 1 H₂SO₄ + 1 Cu(CN)₂

Type of Reaction? ____ Double Replacement

8) $\underline{1}$ GaF₃ + $\underline{3}$ Cs \rightarrow $\underline{3}$ CsF + $\underline{1}$ Ga

Type of Reaction? Single Replacement

Naming Chemical Reactions Practice

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Naming Chemical Reactions Practice Introduction

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