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Chemistry
 Chemistry
 Chemistry

02 E I

Two hours
 Two hours
 Two hours

Instructions:

- Periodic Table is provided.
- This paper consists of 09 pages.
- Answer all the questions.
- Use of calculators is not allowed.
- Write your Index Number in the space provided in the answer sheet.
- Follow the instructions given on the back of the answer sheet carefully.
- In each of the questions 1 to 50, pick one of the alternatives from (1), (2), (3), (4), (5) which is correct or most appropriate and mark your response on the answer sheet with a cross (x) in accordance with the instructions given on the back of the answer sheet.

Universal gas constant $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$
 Avogadro constant $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$

Planck's constant $h = 6.626 \times 10^{-34} \text{ J s}$
 Velocity of light $c = 3 \times 10^8 \text{ m s}^{-1}$

- Which of the following electronic transitions belongs to the visible region of the line spectrum of atomic hydrogen? (n = Principal quantum number)
 - $n = 5 \rightarrow n = 3$
 - $n = 4 \rightarrow n = 2$
 - $n = 1 \rightarrow n = 2$
 - $n = 3 \rightarrow n = 1$
 - $n = 2 \rightarrow n = 1$
- Select the incorrect statement.
 - Pauli exclusion principle excludes the possibility of the presence of a third electron in an orbital.
 - In a potassium atom, the number of electrons having quantum numbers n (principal quantum number) = 3 and m_l (magnetic quantum number) = 0 is 4.
 - The effective nuclear charge felt by a valence electron in nitrogen (N), is greater than the effective nuclear charge felt by a valence electron in carbon (C).
 - Among the ions Na^+ , Mg^{2+} , K^+ and Ca^{2+} , the two ions closest to each other in size are K^+ and Mg^{2+} .
 - Electron gain energy of carbon is negative.
- The increasing order of the second ionization energy ($\text{X}^+(\text{g}) \rightarrow \text{X}^{2+}(\text{g}) + e$) of Be, B and O is,
 - $\text{Be} < \text{B} < \text{O}$
 - $\text{Be} < \text{O} < \text{B}$
 - $\text{B} < \text{O} < \text{Be}$
 - $\text{B} < \text{Be} < \text{O}$
 - $\text{O} < \text{Be} < \text{B}$
- The shapes of F_3ClO , FCIO_2 and FCIO_3 are respectively,
 - tetrahedral, trigonal planar and see-saw.
 - square planar, trigonal planar and tetrahedral.
 - see-saw, trigonal pyramidal and square planar.
 - tetrahedral, trigonal pyramidal and see-saw.
 - see-saw, trigonal pyramidal and tetrahedral.

[See page two]

Gde Chemistry 20memo

Karin Nielsen-Saines



Gde Chemistry 20memo:

The Enigmatic Realm of **Gde Chemistry 20memo**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Gde Chemistry 20memo** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

https://staging.conocer.cide.edu/results/browse/Documents/Laplace_Transforms_And_Control_Systems_For_Technology.pdf

Table of Contents **Gde Chemistry 20memo**

1. Understanding the eBook **Gde Chemistry 20memo**
 - The Rise of Digital Reading **Gde Chemistry 20memo**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Gde Chemistry 20memo**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Gde Chemistry 20memo**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Gde Chemistry 20memo**
 - Personalized Recommendations
 - **Gde Chemistry 20memo** User Reviews and Ratings
 - **Gde Chemistry 20memo** and Bestseller Lists

5. Accessing Gde Chemistry 20memo Free and Paid eBooks
 - Gde Chemistry 20memo Public Domain eBooks
 - Gde Chemistry 20memo eBook Subscription Services
 - Gde Chemistry 20memo Budget-Friendly Options
6. Navigating Gde Chemistry 20memo eBook Formats
 - ePub, PDF, MOBI, and More
 - Gde Chemistry 20memo Compatibility with Devices
 - Gde Chemistry 20memo Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gde Chemistry 20memo
 - Highlighting and Note-Taking Gde Chemistry 20memo
 - Interactive Elements Gde Chemistry 20memo
8. Staying Engaged with Gde Chemistry 20memo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gde Chemistry 20memo
9. Balancing eBooks and Physical Books Gde Chemistry 20memo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gde Chemistry 20memo
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gde Chemistry 20memo
 - Setting Reading Goals Gde Chemistry 20memo
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gde Chemistry 20memo
 - Fact-Checking eBook Content of Gde Chemistry 20memo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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