

QUESTION 2

| | Test | Observation | Inference |
|--------|---|---|-----------------------------------|
| (a)(i) | C + Conc. H_2SO_4 + strong heat | A copious brown irritating fume which turned moist blue litmus paper to red | Gas is NO_2 NO_3^- present |
| (b)(i) | C + water + NaOH in drops, in excess | white gelatinous precipitate precipitate dissolved | Al^{3+} or Zn^{2+} |
| (ii) | C + water + NH_3 in drops, in excess | white gelatinous precipitate precipitate dissolved | Al^{3+} or Zn^{2+} |
| (iii) | C + water + freshly prepared $FeSO_4$ + Conc. H_2SO_4 | A brown ring is formed at the junction of the acid and aqueous solution. | NO_3^- |
| (c)(i) | D + water | D dissolved | soluble |
| (ii) | Solution D + dil HCl | Colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black | H_2S present |
| (iii) | Solution D + NaOH in drops, in excess | white precipitate precipitate insoluble | Ca^{2+} or Pb^{2+} |
| (iv) | solution D + $(NH_4)_2CO_3$ | white precipitate | Ca^{2+} |

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Vijay Pandit



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