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REPUBLIC OF SOUTH AFRICA

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2021**



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# Gauteng Trial Pphysical Science 2014 Memo

**Michal Rosen-Zvi**



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expressed in grams they are still correct because 56

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 $2S + 3O_2 \rightarrow 2SO_3$  is this endothermic or exothermic?

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web  $q = 140 \text{ g} \times 4.184 \text{ J/g} \times 16.0^\circ\text{C} = 9372 \text{ J}$   $35.5 \text{ g} \times 1 \text{ mole KCN} \times 0.5453 \text{ moles} \times 65.1^\circ\text{C} = 1200 \text{ J}$  when a 15 g sample of  $SO_2$  dissolves in 80 g of water in a calorimeter the temperature drops from  $42^\circ\text{C}$  to  $15^\circ\text{C}$  calculate the  $\Delta H$  for the process

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