

100) Carbon disulfide consists of 68.75% carbon. What is the molecular formula of this compound? (100)

$$\text{C} = \frac{68.75}{12} = 5.73 \quad \text{S} = \frac{31.25}{32} = 0.976 \quad \frac{5.73}{0.976} = 5.86 \approx 6$$

$$\text{C}_6\text{S}_6$$

answer is C_6S_6 (100)

101) There are 3 monatomic gases in each buffer are not the (Heads).

102) Describe how each of the following gases are used.

Helium is used for balloons and as a gas in light bulbs.
Neon is used in neon signs and as a gas in light bulbs.
Argon is used in light bulbs and as a gas in light bulbs.

It is used in light bulbs and as a gas in light bulbs.
It is used in light bulbs and as a gas in light bulbs. (100)

103) Describe how each of the following gases are used.

Helium is used for balloons and as a gas in light bulbs.
Neon is used in neon signs and as a gas in light bulbs.
Argon is used in light bulbs and as a gas in light bulbs. (100)

104) Carbon disulfide is a compound of carbon and sulfur. Describe its properties.

Describe each of the following: (100)

What is the molecular formula of carbon disulfide?

answer is CS_2 (100)



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