

Name \_\_\_\_\_

Chemistry Worksheet  
Naming & Formula Writing (Ionic)

Instructions: Write the formulas &/or the names for the compounds listed below

1. Sodium nitrate \_\_\_\_\_
2. Calcium carbonate \_\_\_\_\_
3. Magnesium oxide \_\_\_\_\_
4. Ammonium sulfide \_\_\_\_\_
5. Lead (II) sulfate \_\_\_\_\_
6. Sodium cyanide \_\_\_\_\_
7. Potassium hydroxide \_\_\_\_\_
8. Silver chloride \_\_\_\_\_
9. Iron (III) hydroxide \_\_\_\_\_
10. Potassium hydroxide \_\_\_\_\_
11. Tin (IV) perchlorate \_\_\_\_\_
12. Potassium carbonate \_\_\_\_\_
13. Silver nitrate \_\_\_\_\_
14. Sodium iodide \_\_\_\_\_
15. Ammonium hydroxide \_\_\_\_\_
16. Potassium iodate \_\_\_\_\_
17. Lead (IV) oxide \_\_\_\_\_
18. Ammonium hydroxide \_\_\_\_\_
19. Barium sulfate \_\_\_\_\_
20. barium chloride \_\_\_\_\_
21. Cobalt (II) chloride \_\_\_\_\_
22. Sodium carbonate \_\_\_\_\_
23. Calcium oxide \_\_\_\_\_
24. Lead (II) nitrate \_\_\_\_\_
25. Tin (II) chloride \_\_\_\_\_

26. Aluminum chloride \_\_\_\_\_
27. Iron (III) hydroxide \_\_\_\_\_
28. Sodium acetate \_\_\_\_\_
29. calcium hydroxide \_\_\_\_\_
30. sodium iodate \_\_\_\_\_
31. Nickel (II) nitrate \_\_\_\_\_
32. Iron (II) chloride \_\_\_\_\_
33. Magnesium bromide \_\_\_\_\_
34. Ammonium nitrate \_\_\_\_\_
35. Silver bromide \_\_\_\_\_
36.  $\text{Al}(\text{OH})_3$  \_\_\_\_\_
37.  $\text{NH}_4\text{I}$  \_\_\_\_\_
38.  $\text{Li}_2\text{CO}_3$  \_\_\_\_\_
39.  $\text{CuSO}_4$  \_\_\_\_\_
40. KCN \_\_\_\_\_
41.  $\text{Pb}(\text{ClO})_2$  \_\_\_\_\_
42. BaS \_\_\_\_\_
43.  $\text{ZnSO}_4$  \_\_\_\_\_
44.  $\text{Pb}(\text{CH}_3\text{COO})_2$  \_\_\_\_\_
45.  $\text{Ca}(\text{NO}_3)_2$  \_\_\_\_\_
46.  $\text{Fe}_2(\text{CO}_3)_3$  \_\_\_\_\_
47.  $\text{NH}_4\text{IO}_3$  \_\_\_\_\_
48.  $\text{CaCl}_2$  \_\_\_\_\_
49. NaF \_\_\_\_\_
50.  $\text{Cu}(\text{NO}_3)_2$  \_\_\_\_\_

# Episode 6chemical Formula Answers

**Tao Wei**



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