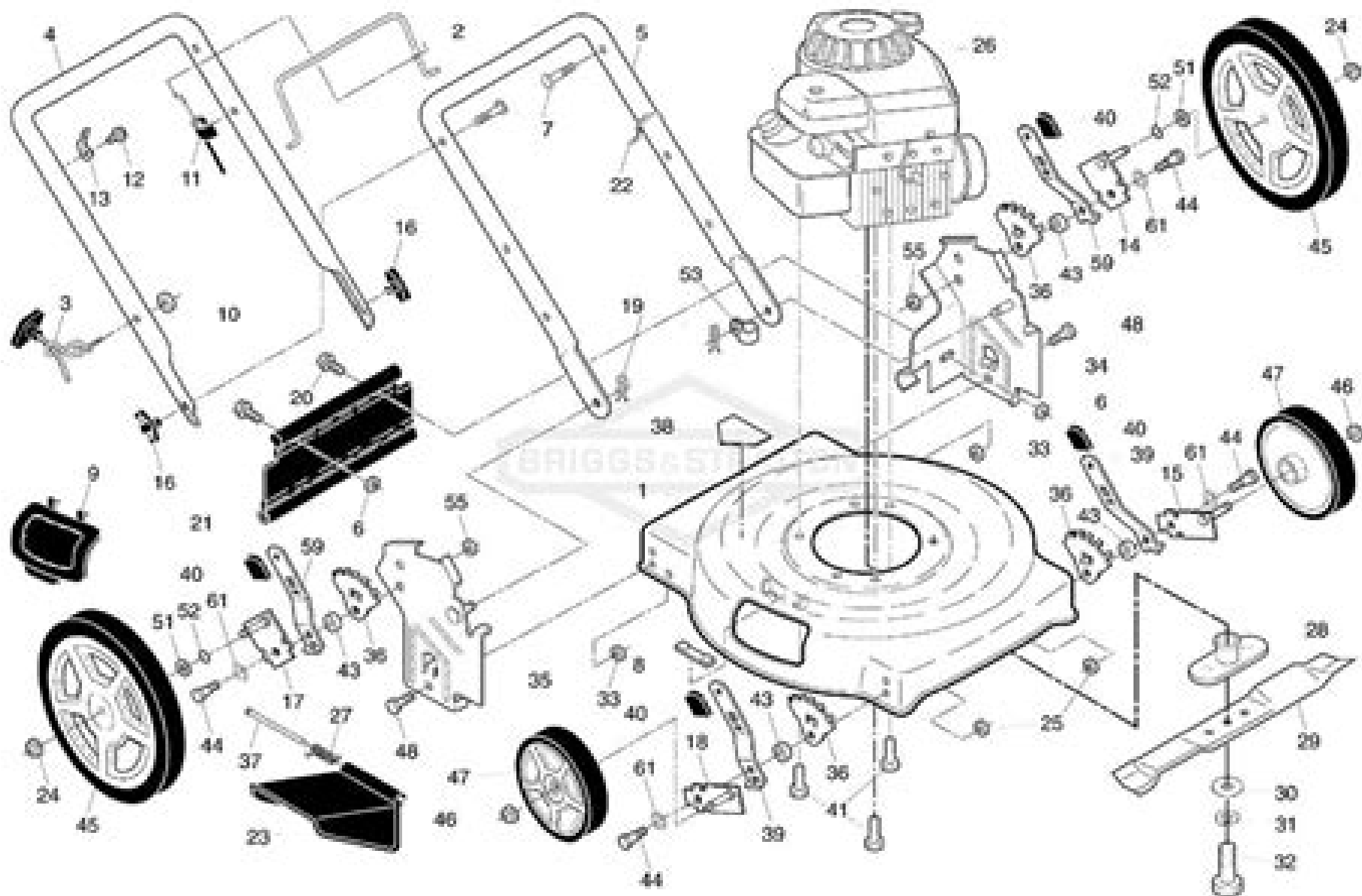




Murray Lawn Mower Tecumseh Engine

Sebastian Brünink

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web the chemical equation used was $\text{Ca(OH)}_2 \text{ s} \rightleftharpoons \text{Ca}^{2+} \text{ aq} + 2\text{OH}^- \text{ aq}$ the solubility product was determined to be $K_{\text{sp}} \text{ Ca(OH)}_2$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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