



| | | | |
|--|---|---|--|
| C_2H_4 $8+4$ # of electrons = <u>12</u> $\text{H}:\ddot{\text{C}}::\ddot{\text{C}}:\text{H}$ | C_2F_4 $8+28$ # of electrons = <u>36</u> $\begin{array}{cc} :\ddot{\text{F}}: & :\ddot{\text{F}}: \\ :\ddot{\text{C}}::\ddot{\text{C}}: \\ :\ddot{\text{F}}: & :\ddot{\text{F}}: \end{array}$ | CO $4+6$ # of electrons = <u>10</u> $:\text{C}:::\text{O}:$ | O_2 $6+6$ # of electrons = <u>12</u> $\ddot{\text{O}}::\ddot{\text{O}}$ |
| CO_2 $4+6+6$ # of electrons = <u>16</u> $\ddot{\text{O}}::\text{C}::\ddot{\text{O}}$ | C_2H_2 (H C C H) $8+2$ # of electrons = <u>10</u> $\text{H}:\text{C}::\text{C}:\text{H}$ | N_2 $5+5$ # of electrons = <u>10</u> $:\text{N}:::\text{N}:$ | HCN $1+4+5$ # of electrons = <u>10</u> $\text{H}:\text{C}:::\text{N}:$ |
| CN^- $4+5+1$ # of electrons = <u>10</u> $[:\text{C}:::\text{N}:]^-$ | SO_4^{2-} $6+24+2$ # of electrons = <u>32</u> $\left[\begin{array}{c} :\ddot{\text{O}}: \\ :\ddot{\text{O}}:\ddot{\text{S}}:\ddot{\text{O}}: \\ :\ddot{\text{O}}: \end{array} \right]^{2-}$ | PO_4^{3-} $5+24+3$ # of electrons = <u>32</u> $\left[\begin{array}{c} :\ddot{\text{O}}: \\ :\ddot{\text{O}}:\ddot{\text{P}}:\ddot{\text{O}}: \\ :\ddot{\text{O}}: \end{array} \right]^{3-}$ | ClO_3^- $7+18+1$ # of electrons = <u>26</u> $\left[\begin{array}{c} :\ddot{\text{O}}: \\ :\ddot{\text{O}}:\ddot{\text{Cl}}:\ddot{\text{O}}: \end{array} \right]^-$ |
| CO_3^{2-} $4+18+2$ # of electrons = <u>24</u> $\left[\begin{array}{c} :\ddot{\text{O}}: \\ :\ddot{\text{O}}:\ddot{\text{C}}:\ddot{\text{O}}: \end{array} \right]^{2-}$  | NO_3^- $5+18+1$ # of electrons = <u>24</u> $\left[\begin{array}{c} :\ddot{\text{O}}: \\ :\ddot{\text{O}}:\ddot{\text{N}}:\ddot{\text{O}}: \end{array} \right]^-$  | SO_2 $6+12$ # of electrons = <u>18</u> $\begin{array}{c} :\ddot{\text{O}}:\ddot{\text{S}}::\ddot{\text{O}}: \\ \updownarrow \\ \ddot{\text{O}}::\ddot{\text{S}}::\ddot{\text{O}}: \end{array}$ | O_3 (O O O) # of electrons = <u>18</u> $\begin{array}{c} \ddot{\text{O}}::\ddot{\text{O}}::\ddot{\text{O}}: \\ \updownarrow \\ :\ddot{\text{O}}::\ddot{\text{O}}::\ddot{\text{O}}: \end{array}$ |
| SF_6 $6+42$ # of electrons = <u>48</u> $\begin{array}{ccccc} & :\ddot{\text{F}}: & & :\ddot{\text{F}}: & \\ :\ddot{\text{F}}: & & :\ddot{\text{S}}: & & :\ddot{\text{F}}: \\ & :\ddot{\text{F}}: & & :\ddot{\text{F}}: & \end{array}$ | XeF_4 $8+28$ # of electrons = <u>36</u> $\begin{array}{ccc} :\ddot{\text{F}}: & & :\ddot{\text{F}}: \\ :\ddot{\text{F}}:\ddot{\text{Xe}}:\ddot{\text{F}}: & & \\ :\ddot{\text{F}}: & & :\ddot{\text{F}}: \end{array}$ | PCl_5 $5+35$ # of electrons = <u>40</u> $\begin{array}{ccc} :\ddot{\text{Cl}}: & & :\ddot{\text{Cl}}: \\ :\ddot{\text{Cl}}:\ddot{\text{P}}:\ddot{\text{Cl}}: & & \\ :\ddot{\text{Cl}}: & & :\ddot{\text{Cl}}: \end{array}$ | SeF_4 $6+28$ # of electrons = <u>34</u> $\begin{array}{ccc} :\ddot{\text{F}}: & & :\ddot{\text{F}}: \\ :\ddot{\text{F}}:\ddot{\text{Se}}:\ddot{\text{F}}: & & \\ :\ddot{\text{F}}: & & :\ddot{\text{F}}: \end{array}$ |

Molecules Lewis Dot Structure Answer Key

Denise Guinn, Rebecca Brewer



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