

1. Calculate the molar mass of the compound (just answer in scientific notation)

$$C_2H_5NO_2 \rightarrow \frac{1 \text{ mol } C_2H_5NO_2}{1 \text{ mol } C_2H_5NO_2} \times \frac{1 \text{ mol } C_2H_5NO_2}{1 \text{ mol } C_2H_5NO_2} = 75.07 \text{ g/mol}$$

2. Calculate the mass of the compound.

$$52.3 \text{ g} \times \frac{1 \text{ mol } C_2H_5NO_2}{75.07 \text{ g/mol}} = 0.696 \text{ mol}$$

3. Calculate the volume of the compound.

$$0.696 \text{ mol} \times \frac{1 \text{ mol } C_2H_5NO_2}{1 \text{ mol } C_2H_5NO_2} = 0.696 \text{ mol}$$

4. Calculate using molar mass (just answer in scientific notation)

$$C_2H_5NO_2 \rightarrow \frac{1 \text{ mol } C_2H_5NO_2}{75.07 \text{ g/mol}} \times \frac{1 \text{ mol } C_2H_5NO_2}{75.07 \text{ g/mol}} = 4.5 \times 10^{-2} \text{ mol}$$

5. What is the empirical formula of the compound?



6. What is the empirical formula of the compound?



7. What is the empirical formula of the compound?



8. Name the compound (just answer in scientific notation)



9. Name the compound (just answer in scientific notation)



10. Name the compound (just answer in scientific notation)



11. Name the compound (just answer in scientific notation)



12. Name the compound (just answer in scientific notation)



13. Name the compound (just answer in scientific notation)



14. Name the compound (just answer in scientific notation)



15. Name the compound (just answer in scientific notation)



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Herman Joseph Heuser



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