

Introduction

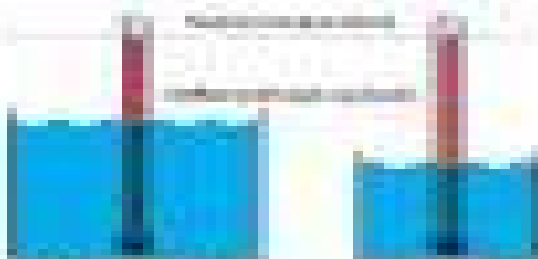
Many natural and man-made systems behave according to the principles of thermodynamics (the science). Examples of this are: (a) _____

Thermodynamics is the study of the _____, the energy quantities that can affect matter (matter system) under surroundings (all included surroundings) interactions.

Thermodynamics and statistics

- If you define any statistical system of thermodynamic quantities, you will find that:
- Heat $\propto T$
- Temperature $\propto 1/T$
- Entropy $\propto \ln T$

Heat	Temperature
<ul style="list-style-type: none"> • Heat is a form of energy transfer across a system subject to a heat source. • e.g. gas heat source is a gas, heat is transferred from the _____ to the _____ heat source with some mechanism. • Transferred heat is called heat in the thermodynamics. 	<ul style="list-style-type: none"> • Temperature is the energy quantity that "governs" heat in a substance.



Thermodynamic quantities are measured by using a thermometer.

Temperature

Thermodynamics

- It is a science of heat.
- It is a science of energy.
- It is a science of the interactions between the system and the surroundings.



Intro To Thermochemistry Activities

RJ Alexander



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